

A New Dimeric Sesquiterpenoid from *Chloranthus japonicus* Sieb.

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Abstract: Phytochemical investigation into the whole plants of *Chloranthus japonicus* Sieb. led to isolation and identification of a new dimeric lindenane sesquiterpenoid, named chlojapolactone B (**1**), and two new phenolic derivatives (**2** and **3**). Their chemical structures were elucidated on the basis of HRESIMS and NMR spectroscopic data, and the absolute configuration of **1** was determined using the electronic circular dichroism (ECD) analysis.

Keywords: *Chloranthus japonicus*; dimeric sesquiterpenoid; phenolic derivative. © 2019 ACG Publications. All rights reserved.

1. Introduction

Chloranthus japonicus Sieb., (Family: Chloranthaceae), is a perennial herb widely distributed in East Asia, including China, Japan, and Korea. In Chinese folklore medicines, this plant is commonly used for the treatment of traumatic injury, rheumatic arthralgia, fracture, pulmonary tuberculosis, and neurasthenia [1]. Previous phytochemical investigations have demonstrated that *C. japonicus* is enriched with sesquiterpenoids and sesquiterpenoid dimers [2–10]. A variety of sesquiterpenoids including eudesmane-, lindenane-, germacrane-, and acorane-, as well as lindenane-type sesquiterpenoid dimers and trimers with various bioactivities have been reported from this plant [11–16]. Among them, dimers of lindenane-type sesquiterpenoid showed more significant bioactivities such as anti-HIV-1, anti-HCV, activated AMPK effects, anti-inflammatory, anti-tumor and DNA Topoisomerase I inhibitory activities [3, 9, 15–18]. In our phytochemical investigation on this plant for structurally intriguing and biologically important compounds, a new dimeric sesquiterpenoid (**1**) and two new phenolic derivatives (**2** and **3**) were obtained and characterized from the whole plants of *C. japonicus*. Herein, the isolation and structural elucidation of these compounds, together with their bioactivity evaluation, are described (Figure 1).

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Qi-Yan Li and Yan Wang contributed equally to this work.

2. Materials and Methods

2.1. Material

The whole plants of *C. japonicus* were collected in October 2016, from Jilin province of China. This plant material was identified by Prof. Peiming Yang, China State Institute of Pharmaceutical Industry. A voucher specimen (No. 20161012) was deposited in Shanghai University of Medicine & Health Sciences Central Laboratory. ^1H (600 MHz) and ^{13}C (150 MHz) NMR spectra were determined using a Bruker AscendTM 600 spectrometer. HRESIMS data were performed using an Agilent Technologies 6230 Accurate Mass Q-ToF UHPLC/MS spectrometer. Silica gel (200–300 mesh) for column chromatography and GF254 for TLC were purchased from Qingdao Haiyang Chemical Co. Ltd. (Qingdao, China). Optical rotations were measured using a Rudolph Research Autopol I automatic polarimeter. UV and electronic circular dichroism (ECD) spectra were recorded on a JASCO High Performance J-1500 CD spectrometer.

2.2. Extraction and Isolation

The whole plants of *C. japonicus* (5.0 kg) were dried, powdered, and subsequently extracted three times with 80% EtOH under continuous reflux. The filtrate was concentrated to obtain a crude extract, which was dissolved in water and then extracted with EtOAc and *n*-BuOH. The EtOAc fraction (112.5 g) was subjected to silica gel column chromatography (CC) and eluted with a gradient of increasing acetone (0–100%) in petroleum to afford six fractions (Fr.1–Fr.6). Fraction Fr. 2 (10.5 g) was subjected to CC over RP18 gel and eluted with MeOH-H₂O (35:65–100:0) to give four sub-fractions (Fr.2-1–Fr.2-4). Fr.2-2 was initially eluted with MeCN-H₂O (50:50) on an RP C8 column and further purified by semipreparative HPLC with MeOH-H₂O (40:60) elution to afford compound **1** (2.0 mg). Fr. 4 (8.0 g) was subjected to CC over RP C18 and eluted with MeOH-H₂O in a stepped gradient (30:70–100:0) to afford five sub-fractions (Fr.4-1–Fr.4-5). Fr. 4-3 was separated by semipreparative HPLC eluted with MeOH-H₂O (55:45) to afford compounds **2** (4.0 mg) and **3** (5.0 mg).

2.3 Spectroscopic Data

Compound 1: White amorphous powder. $[\alpha]_{\text{D}}^{22.5} -70.0$ (*c* 0.3, MeOH); UV (MeOH). λ_{max} (log ϵ): 200 (4.18), 290 (3.40); ECD (MeOH): 200 ($\Delta\epsilon +2.53$), 215 ($\Delta\epsilon -4.08$) nm; IR, ν_{max} 3450, 2361, 1643, 671 cm^{-1} ; ^1H (600 MHz, CDCl_3) and ^{13}C (150 MHz, CDCl_3) NMR data: Table 1; HRESIMS m/z 527.2048 $[\text{M} + \text{Na}]^+$ (calcd for $\text{C}_{30}\text{H}_{32}\text{O}_7\text{Na}$ 527.2156), 522.2485 $[\text{M} + \text{NH}_4]^+$ (calcd for $\text{C}_{30}\text{H}_{32}\text{O}_7\text{NH}_4$ 522.2486).

Compound 2: Yellow oil. $[\alpha]_{\text{D}}^{22.5} -25.6$ (*c* 0.3, MeOH); λ_{max} (log ϵ): 200 (4.08), 232 (3.53); IR, ν_{max} 3741, 2985, 2361, 1652, 1563, 1464, 1183 cm^{-1} ; ^1H (600 MHz, CDCl_3) and ^{13}C (150 MHz, CDCl_3) NMR data: Table 2; HRESIMS m/z 355.1875 $[\text{M} + \text{Na}]^+$ (calcd for $\text{C}_{20}\text{H}_{28}\text{O}_4\text{Na}$ 355.1880), 333.3063 $[\text{M} + \text{H}]^+$ (calcd for $\text{C}_{20}\text{H}_{29}\text{O}_4$ 333.2060).

Compound 3: Yellow oil. $[\alpha]_{\text{D}}^{22.5} -21.7$ (*c* 0.3, MeOH); λ_{max} (log ϵ): 200 (4.15), 232 (3.46); IR, ν_{max} 3744, 2987, 2362, 1739, 1542, 1276, 685 cm^{-1} ; ^1H (600 MHz, CDCl_3) and ^{13}C (150 MHz, CDCl_3) NMR data: Table 2; HRESIMS m/z 397.1979 $[\text{M} + \text{Na}]^+$ (calcd for $\text{C}_{22}\text{H}_{30}\text{O}_5\text{Na}$ 397.1985), 375.2158 $[\text{M} + \text{H}]^+$ (calcd for $\text{C}_{22}\text{H}_{31}\text{O}_5$ 375.2166).

2.4. ECD Calculation Method

Monte Carlo conformational study was carried out by means of the Spartan's 10 software using Merck Molecular Force Field (MMFF). The conformers with Boltzmann-population of over 1% were chosen for ECD calculations, and then the conformers were initially optimized at B3LYP/6-31+g (d, p) level in methanol using the CPCM polarizable conductor calculation model. The theoretical calculations of ECD were conducted in methanol using time-dependent density functional theory (TD-

DFT) at the B3LYP/6-31+g (d, p) level for all conformers of the compounds. Rotatory strengths for a total of 30 excited states were calculated. ECD spectra were generated using the program SpecDis 1.6 (University of Würzburg, Würzburg, Germany) and GraphPad Prism 5 (University of California San Diego, USA) from dipole-length rotational strengths by applying Gau6-1-Rian band shapes with $\sigma = 0.3$ eV.

2.5. Inhibitory Effect Against LPS-induced Tumor Necrosis Factor (TNF)- α in RAW264.7 Macrophages

Briefly, the RAW264.7 cell line was cultured in RPMI medium supplemented with 0.1% sodium bicarbonate and 2 mM glutamine, penicillin G (100 units/mL), streptomycin (100 $\mu\text{g/mL}$) and 10% fetal calf serum. The cells were harvested with trypsin-EDTA and diluted to a suspension in a fresh medium. The cells were seeded in the 96-well plates containing 1.0×10^5 cells/well and allowed to adhere for 1 h at 37°C under a humidified atmosphere containing 5% CO_2 . Then, the medium was replaced with a fresh medium containing 100 $\mu\text{g/mL}$ of LPS together with the compounds at various concentrations and then incubated for 48 h. The supernatant was transferred into the 96-well ELISA plates and then TNF- α concentrations were determined using commercial ELISA kits. The test compounds were dissolved in DMSO, and the solution was added to RPMI.

3. Results and Discussion

3.1. Structure Elucidation

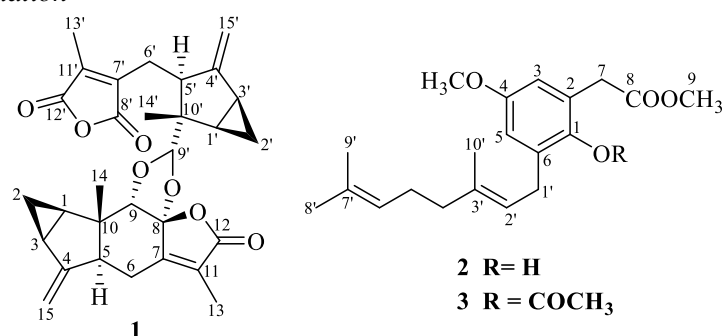


Figure 1. Chemical structures of compounds 1–3

Table 1. ^1H (600 MHz) and ^{13}C (150 MHz) NMR data for compound 1 in CDCl_3

Pos.	δ_{C}	δ_{H} (mult, J in Hz)	Pos.	δ_{C}	δ_{H} (mult, J in Hz)
1	23.5	1.84 (m)	1'	28.2	1.85 (m)
2	15.7	0.85(m), 0.67 (m)	2'	11.2	0.81(m), 0.70 (m)
3	23.7	2.00 (m)	3'	25.0	1.89 (m)
4	151.8		4'	153.5	
5	50.4	3.42 (m)	5'	44.4	2.75 (dd, 12.0, 5.5)
6	22.3	2.46 (m)	6'	28.1	2.52 (ddd, 13.2, 5.5, 1.0)
		2.21 (d, 13.5)			2.25 (d, 12.0)
7	154.6		7'	143.6	
8	109.7		8'	165.8	
9	85.4	4.26 (s)	9'	109.4	5.08 (s)
10	42.8		10'	48.1	
11	127.6		11'	142.0	
12	171.4		12'	166.2	
13	8.8	1.84 (d, 1.3)	13'	10.3	2.01 (s)
14	17.3	0.47 (s)	14'	15.5	1.22 (s)
15	106.5	5.02 (brs), 4.71 (brs)	15'	107.6	4.89 (brs), 4.33 (brs)

Chlojapolactone B (**1**) was isolated as a white amorphous powder. The molecular formula of $C_{30}H_{32}O_7$ was established by HR-ESIMS data and ^{13}C NMR data (Table 1). HRESIMS data displayed a quasi-molecular ion peak at m/z 522.2485 $[M+NH_4]^+$ (calcd for 522.2486), the molecular formula contained 15 degrees of unsaturation. The IR spectrum exhibited characteristic absorption bands of carbonyl (1643 cm^{-1}) groups. In accordance with the molecular formula, 30 carbon resonances consisting of three carbonyls, four double bonds (two tetrasubstituted and two exocyclic), four methylenes, eight methines (two oxygenated), three quaternary carbons (one oxygenated), and four methyls were resolved in the ^{13}C NMR and categorized by DEPT experiments. The 1H NMR spectroscopic data (Table 1) showed two pairs of terminal double bonds (δ_H 5.02, 4.71, 4.89, 4.33), four methyl groups (δ_H 1.84, 0.47, 2.01, 1.22), two oxygenated methines (δ_H 5.08 and 4.26), and four highly upfield-shifted protons (δ_H 0.85, 0.67, 0.81, 0.70). The 1H - 1H COSY spectrum showed two sets of proton spin systems in a 1,2-distributed cyclopropane ring (δ_H 0.67, 0.85, 1.84, and 2.00; δ_H 0.70, 0.81, 1.85, and 1.89). The above data revealed that compound **1** should be a lindenane sesquiterpenoid dimer [10, 19]. Detailed 2D NMR analysis afforded the basic structure of the two units (unit A and B) as depicted in Figure 2. Unit B was characterized as the typical lindenane-type sesquiterpenoid, chloranthalactone E, by comparing with their NMR data [20]. This was further confirmed by 1H - 1H COSY and HMBC correlations. In unit A, two structural fragments, including cyclopropane ring (C-1'-C-2'-C-3') and a C-5'-C-6' system were established by the 1H - 1H COSY correlations of H-1'/H-2'/H-3' and H-5'/H-6'. The HMBC correlations of H₃-14' to C-1', C-5', C-9' and C-10' allowed to establish the connection of C-1', C-5', C-9' and C-14' to the quaternary carbon C-10'. The linkage of C-6'-C-7'-C-8' was revealed by the HMBC correlations of H-6'/C-7' to C-8', while the HMBC correlations from H₃-13' to C-7', C-11', and C-12', as well as from H-6' to C-7' and C-11' suggested the C-7'-C-11'-C-12' linkage pattern. The upfield-shifted carbonyls at C-8' and C-12' required an oxygen bridge between C-8' and C-12', which eventually formed a 3-methyl-2,5-furandione moiety in unit A. As the above-mentioned structural elucidation already accounted for 14 out of the 15 degrees of unsaturation, the remaining one thus required the presence of an additional ring to link units A and B. In the HMBC spectrum, obvious correlations from H-9' to the ketal carbon at C-8 and C-9, as well as the correlations of H-9/C-9' suggested the presence of two oxygen bridges between C-9' and C-8, C-9' and C-9, respectively. Based on the acetal nature of C-9' and downfield chemical shift of C-8, two oxygen bridges are required to form a 1,3-dioxine ring. Thus, the planar structure of **1** was elucidated as depicted, with a 1,3-dioxolane ring linking two lindenane sesquiterpenoid units.

The relative configuration of **1** was established by NOESY experiment and by comparing the NMR data with those known lindenane monomers. In unit B, the correlations of H-2 β /H₃-14, and H₃-14/H-6 β indicated that these protons were cofacial and arbitrarily assigned as β -orientation, while H-2 α , H-1, H-3, and H-5 were as α -oriented by the NOESY correlations of H-2 α /H-1, H-3 and the large coupling constants between H-5 and H-6 β ($J = 13.0\text{ Hz}$) (Figure 2). The H₃-14 methyl signal at δ_H 0.47 supported *syn*-relationship of the C-14 methyl group and 8-oxygen atom, suggesting C-8-O-C-12 in β -orientation [21, 22]. Meanwhile, the 1D NMR data of the chiral centers in unit B were consistent with those reported in its analogues [10, 23]. As for unit A, the NOESY correlations of H₃-14'/H-2' β indicated that these protons were cofacial and arbitrarily assigned as β -oriented. In consequence, the NOESY correlations of H-5'/H-3' and H-5'/H-1' revealed that these protons were α -oriented. In addition, the NOESY correlations of H-9/H-9' indicated that these protons were co-facial on the 1,3-dioxolane ring. Above mentioned NOESY correlations confirmed the relative configuration of **1** was similar with that of chlojapolactone A [10]. The absolute configuration of compound **1** was determined by comparison the experimental and the computational ECD spectra. As shown in Figure 3, the ECD spectrum of **1** showed a negative Cotton effect at 215 nm ($\pi \rightarrow \pi^*$ transition) due to the transition interaction between the α,β -unsaturated ketone and furan dione chromophores. The calculated ECD curve of enantiomer configuration of (1*R*,3*S*,5*S*,8*S*,9*S*,10*S*,1'*R*,3'*S*,5'*S*,9'*R*,10'*S*)-**1** matched well with the experimental one, assigning unambiguously the absolute configuration of compound **1** as 1*R*,3*S*,5*S*,8*S*,9*S*,10*S*,1'*R*,3'*S*,5'*S*,9'*R*,10'*S*. In addition, from the prospective of biosynthesis pathway, compound **1** could be considered as a precursor of chlojapolactone A. Therefore, this assignment was consistent with the biogenetic origin of lindenane-type sesquiterpenoids from the genus *Chloranthus*. Thus, compound **1** was assigned as depicted.

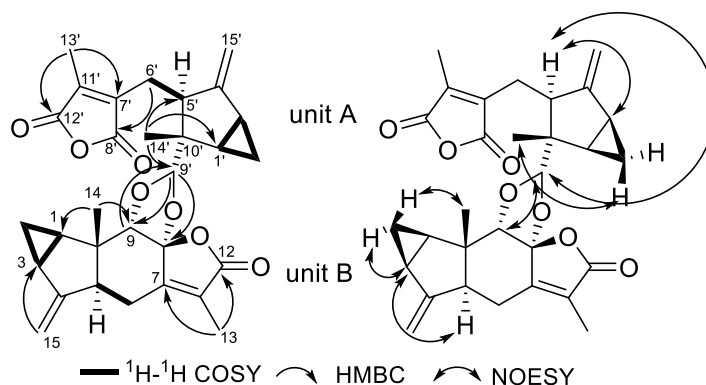


Figure 2. The key ^1H - ^1H COSY, HMBC and NOESY correlations of compound **1**

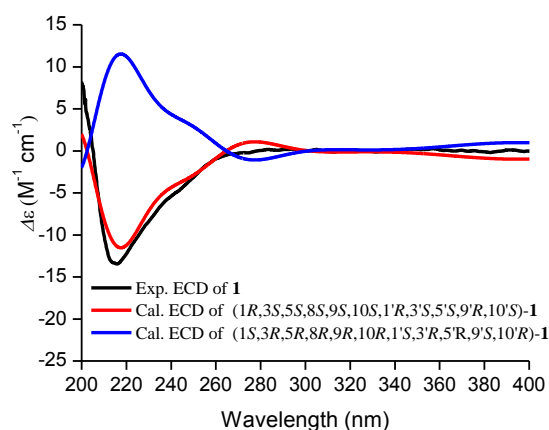


Figure 3. Calculated and experimental ECD spectra of **1**

Compound **2** was assigned to a molecular formula of $\text{C}_{20}\text{H}_{28}\text{O}_4$ by HRESIMS ion at m/z 333.2063 $[\text{M}+\text{H}]^+$ (calcd for 333.2060). A strong absorption at 1652 cm^{-1} in the IR spectrum suggested the presence of a carbonyl group. The ^1H and ^{13}C NMR spectral data of **2** were similar to those of denudaquinol [24], the major difference being the presence of a methoxy group. The key HMBC correlations of $-\text{OCH}_3/\text{C}-4$ suggested that the methoxy group was located at C-4. Therefore, the structure of **2** was characterized to be 4-methoxyl-denudaquinol. Compound **3** was isolated as yellow colored oil. A protonated molecular ion at m/z 375.2158 $[\text{M}+\text{H}]^+$, calcd for 375.2166 indicated a molecular formula of $\text{C}_{22}\text{H}_{30}\text{O}_5$. The ^1H and ^{13}C NMR spectral data of **3** was comparable to that of **2**. The main difference was the presence of an acetyl group. The HMBC correlations of $-\text{CH}_3$ (δ_{H} 2.29) to the carbonyl carbon (δ_{C} 169.5) and C-1 indicated that the acetoxy group was located at the C-1 position. The structure of **3** was established as 1-acetyl-4-methoxyl-denudaquinol. Because methanol and ethanol were used in the extraction and isolation process, compounds **2** and **3** might be artificial products.

Chlojapolactone B (**1**) was evaluated for its inhibitory effect against LPS induced TNF- α production in RAW264.7 macrophages. Compound **1** exhibited moderate inhibition against TNF- α with an IC_{50} value of $76.16\ \mu\text{M}$, which was comparable to the positive control triptolide ($\text{IC}_{50} = 11.50\ \mu\text{M}$).

Table 2. ^1H (600 MHz) and ^{13}C (150 MHz) NMR data for compounds **2** and **3** in CDCl_3

Pos.	2		3	
	δ_{C}	δ_{H} (mult, <i>J</i> in Hz)	δ_{C}	δ_{H} (mult, <i>J</i> in Hz)
1	147.2		141.3	
2	121.7		127.7	
3	113.7	6.54 (d, 3.0)	113.8	6.69 (d, 3.0)
4	153.4		157.4	
5	115.0	6.65 (d, 3.0)	114.7	6.70 (d, 3.0)
6	130.5		135.4	
7	37.6	3.64 (s)	37.0	3.68 (s)
8	174.0		171.3	
9	52.7	3.73 (s)	52.3	3.68 (s)
1'	29.6	3.36 (d, 7.2)	28.9	3.36 (d, 7.2)
2'	121.9	5.31 (brt, 7.2)	121.2	5.23 (brt, 7.2)
3'	138.0		137.6	
4'	39.9	2.07 (m)	39.8	2.04 (m)
5'	26.7	2.11 (m)	26.8	2.09 (m)
6'	124.2	5.09 (brt, 6.7)	124.2	5.10 (brt, 7.0)
7'	131.9		131.8	
8'	17.8	1.60 (s)	17.9	1.60 (s)
9'	25.8	1.68 (s)	25.8	1.68 (s)
10'	16.3	1.74 (s)	16.3	1.67 (s)
OCH ₃	55.8	3.74 (s)	55.6	3.77 (s)
-COCH ₃			169.5	
			20.6	2.29 (s)

4. Conclusion

As part of our ongoing study on the traditional herbal medicines in China. We have investigated the chemical constituents of *C. japonicus* to isolate one new dimeric lindenane sesquiterpenoid and two new phenolic derivatives. Chlojapolactone B (**1**), a novel lindenane sesquiterpenoid dimer featuring a rare 1,3-dioxolane linkage between an 8,9-*seco* lindenane and a lindenane sesquiterpenoid. The planar structures of these compounds were determined by 1D and 2D NMR as well as by HRESIMS spectral analysis. The absolute configuration of **1** was elucidated by ECD spectroscopic analyses. These compounds might contribute the chemical diversity of *C. japonicus*.

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Supporting Information

Supporting information accompanies this paper on <http://www.acgpubs.org/journal/records-of-natural-products>

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