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Antiradical Aromatic Constituents from *Pleurotus eryngii*

Xin-Xin Cao¹, Jin-Yue Sun², Chao Liu², Jun-Sheng Zhang¹^{*} and

Hua Zhang^{1*}

¹School of Biological Science and Technology, University of Jinan, Jinan 250022, China ²Institute of Agro-Food Science and Technology, Shandong Academy of Agricultural Sciences, Jinan 250100, China

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Abstract: Six aromatic compounds including a new (*Z*)-ferulic acid ester (1) and a new syringic acid ester (2), as well as four known ferulic acid esters (3–6), were isolated from a well-known edible mushroom, *Pleurotus eryngii*. Structures of these aromatics were unambiguously characterized via spectroscopic methods especially MS and NMR techniques. All the isolates were assessed for their antiradical effect and NO production inhibitory activity in RAW264.7 macrophages, and compounds 1 and 3 displayed mild DPPH radical removing activity, with IC₅₀ values of 25.0 and 21.6 μ M, respectively.

Keywords: *Pleurotus eryngii*; aromatics; ferulic acid ester; syringic acid ester; antiradical. © 2020 ACG Publications. All rights reserved.

1. Introduction

The king oyster mushroom (*Pleurotus eryngii*) is a well-known edible macro mushroom cultivated all over the world and has been reported to possess various pharmacological benefits to human health [1]. The bioactive constituents of *P. eryngii* have been intensively investigated in recent years. The macromolecule ingredients from this mushroom mainly include proteins [2,3] and polysaccharides [4], and great efforts have been particularly devoted to the study of polysaccharides due to their diverse bioactivities such as antitumor and immunomodulatory activities [4]. In addition to the macromolecules, small molecule natural products have also attracted attentions from the scientific community. Steroids are so far the mostly reported compounds and they show potential NO production and aromatase inhibitory effect [5–9]. Moreover, cytotoxic [10,11], anti-osteoporotic [12], NQO1 induction [13], antioxidant [14] and anti-inflammatory [11] phenolic constituents have also been reported. Other minor constituents [15] comprise cytotoxic triterpenoids [16] and diterpenoids [17], as well as sesquiterpenoids [13,18] and monoterpenoids [19].

As a project to explore the further economic and pharmacological values of bulk edible mushrooms in Shandong province, we have chosen *P. eryngii* as one of our research subjects and carried out an intensive chemical study into the small molecule constituents. As a result, an array of aromatic compounds, including five ferulic acid esters (1 and 3-6) and a syringic acid ester (2), have been obtained and structurally characterized, and the structures of 1 and 2 have been reported for the first time. Our bioassays established that the (*Z*)-ferulic acid esters 1 and 3 exhibited comparable antiradical activity as the positive control ascorbic acid. The isolation, structure identification and biological assessment of the aromatic compounds are thus depicted below.

^{*} Corresponding author: E-Mail: <u>bio_zhangjs@ujn.edu.cn</u>

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Figure 1. Structures of aromatic compounds from P. eryngii

2. Materials and Methods

2.1. General

NMR spectra were acquired from a Bruker Avance DRX600 spectrometer (Bruker BioSpin AG, Fällanden, Switzerland) and the residual solvent peaks (CDCl₃ δ_H 7.26, δ_C 77.16) were used as references. HR-MS analyses were performed on Agilent 6545 Q-TOF MS spectrometer (Agilent Technologies Inc., Waldbronn, Germany) with separate ESI and APCI ion source. The HPLC analyses/separations were carried out on a Shimadzu 20A series LC instrument (Shimadzu, Tokyo, Japan) coupled with a YMC-Pack ODS-A column (5 µm, 10 × 250 mm, YMC Co. Ltd., Tokyo, Japan). D101-macroporous absorption resin (Sinopharm Chemical Reagent Co. Ltd., Shanghai, China), Sephadex LH-20 gel (GE Healthcare Bio-Sciences AB, Uppsala, Sweden), reversed phase (RP) C₁₈ silica gel (Merck KGaA, Darmstadt, Germany) and silica gel (300–400 mesh, Qingdao Marine Chemical Co. Ltd., Qingdao, China) were mainly used for column chromatography (CC). Analytical grade solvents (Tianjin Fuyu Fine Chemical Co. Ltd., Tianjin, China) were used for CC, and chromatographic grade solvents (Oceanpak Alexative Chemical Ltd., Goteborg, Sweden) were used for HPLC.

2.2. Fungal Material

The fresh fungal materials of *Pleurotus eryngii* were bought in May 2019 from the market of Jinan city and were identified by Prof. Guo-hua Ye of Shandong College of Traditional Chinese Medicine. The voucher specimen of the fungus has been deposited at the corresponding author's institution with an access number of npmc-056.

2.3. Extraction and Isolation

The dried fruiting bodies of *P. eryngii* (10.0 kg) were chopped into pieces and then dipped in 95% EtOH at r.t. for three times (once a week). The EtOH extraction (700 g) was combined and evaporated in vacuum to give a brown residue, which was then partitioned between EtOAc (1.0 L × 3) and water (1.5 L). The EtOAc layer was combined with solvent being removed *in vacuo* to afford the EtOAc partition (260 g). The EtOAc part was fractionated by D-101 macroporous resin CC (EtOH-H₂O, 30%, 50%, 70% and 95%) to obtain four elutions. The 70% ethanol elution (56 g) was fractionated *via* silica gel CC using petroleum ether (PE)-EtOAc (99:1 to 80:20) to afford eight fractions (Frs. 1–8). Fr. 5 was subject to a RP-18 column (70% to 100% MeOH-H₂O) to obtain five sub-fractions (Frs. 5A–5J), and Fr. 5A was further purified by semi-preparative HPLC (3.00 mL/min, 85% MeCN-H₂O) to yield **3** (1.2 mg, $t_R = 18.2 \text{ min}$). Fr. 5B was chromatographed on RP-18 CC again, with 60% to 100% MeOH-H₂O as eluent, to afford two sub-fractions, and the second one was purified by semi-preparative HPLC (3.00 mL/min, 80% MeCN-H₂O) to afford compounds **4** (21.3 mg, $t_R = 18.2 \text{ min}$).

29.0 min), **5** (37.0 mg, $t_R = 28.0$ min) and **6** (30.7 mg, $t_R = 17.0$ min). Fr. 6 was separated by Sephadex LH-20 CC (in 1:1 CH₂Cl₂-MeOH) to return six fractions (Frs. 6A–6F), and Fr. 6F was then subject to RP-18 CC (70% to 100% MeOH-H₂O) to afford four sub-fractions (Frs. 6A1–6A4). Fraction 6A2 was first separated by silica gel CC using CH₂Cl₂-MeOH as eluent (200:1 to 60:1) and then purified by HPLC (3.00 mL/min, 90% MeCN-H₂O) to yield compound **1** (4.7 mg, $t_R = 13.5$ min). Fr. 6A3 was first submitted to Sephadex LH-20 CC (in 1:1 CH₂Cl₂-MeOH) and then purified by semi-preparative HPLC (3.00 mL/min, 90% MeCN-H₂O) to yield compound **2** (2.5 mg, $t_R = 16.0$ min).

Compound 1: Colorless gum; UV (MeOH) λ_{max} (log ε) 307 (4.48); ¹H and ¹³C NMR data (CDCl₃) see Table 1; (+)-HR-ESIMS *m*/*z* 265.1432 [M + H]⁺ (calcd for C₁₅H₂₁O₄⁺, 265.1434).

Compound **2**: Colorless gum; UV (MeOH) λ_{max} (log ε) 275 (4.45); ¹H and ¹³C NMR data (CDCl₃) see Table 1; (+)-HR-APCIMS *m/z* 281.1388 [M + H]⁺ (calcd for C₁₅H₂₁O₅⁺, 281.1384).

2.4 Bioassays

2.4.1. DPPH Radical Scavenging Aactivity

The antioxidant property of all these compounds was assessed by the DPPH radical removing assay as we described previously [20-22] with ascorbic acid as the positive control.

2.4.2 NO Production Inhibitory Assay

The nitric oxide (NO) production inhibitory capability of these aromatics was assessed in a lipopolysaccharide (LPS)-induced murine RAW264.7 macrophage model as we described formerly [23].

3. Results and Discussion

Compound **1** was isolated as a colorless gum, with the molecular formula of $C_{15}H_{20}O_4$ as determined by the (+)-HR-ESIMS analysis at m/z 265.1432 (calcd for $[M + H]^+$, 265.1434). Apart from the resonances [δ_H 7.76 (d, J = 1.9 Hz), 7.10 (dd, J = 8.1, 1.9 Hz), 6.90 (d, J = 8.1 Hz); 6.79, 5.82 (both d, J = 12.8 Hz); 5.83 (s, OH); 3.93 (s, 3H)] observed for a Z-feruloyl unit as in methyl (Z)-ferulic acid ester (**3**) [22], the ¹H NMR data (Table 1) for **1** also revealed the existence of an amyloxy group at δ_H 4.12 (t, J = 6.8 Hz, 2H), 1.65 (m, 2H), 1.33 (m, 4H) and 0.90 (t, J = 6.8 Hz, 3H). Analysis of the ¹³C NMR data (Table 1) also confirmed the aforementioned assignments with typical signals for the Z-feruloyl and amyloxy fragments, and the two fragments were connected *via* an ester bond as evidenced by the chemical shift of H₂-1' (δ_H 4.12) and its HMBC correlation (Fig. 2) to C-9 (δ_C 166.8). Compound **1** was thus elucidated to be amyl (Z)-ferulic acid ester.



Figure 2. ¹H–¹H COSY and selected HMBC correlations for 1 and 2

Compound **2** was obtained as a colorless gum with the molecular formula of $C_{15}H_{20}O_5$ as deduced from the (+)-HR-APCIMS ion at m/z 281.1388 (calcd for $[M + H]^+$, 281.1384). Examination of the NMR data (Table 1) for **2** established the existence of a syringoyl residue at δ_H 7.32 (s, 2H) and δ_C 166.5, 146.7 (2C), 139.3, 121.5, 106.7 (2C), as well as the existence of a (*Z*)-3-hexenyloxy unit at δ_H 5.54 (dtt, J = 10.7, 7.3, 1.5 Hz), 5.40 (dtt, J = 10.7, 7.0, 1.5 Hz), 4.30 (t, J = 6.9 Hz, 2H), 2.52 (m, 2H), 2.10 (m, 2H), 0.97 (t, J = 7.5 Hz), and δ_C 134.8, 123.9, 64.6, 27.1, 20.8, 14.4. Further

interpretation of 2D NMR data (Fig. 2) corroborated the aforementioned assignments, revealing key ¹H-¹H COSY fragments of H₂-1' to H₃-6' and HMBC correlations from H-2(6) to C-1, C-3(5), C-4 and C-7. Finally, the two structural fragments were linked via an ester bond as demonstrated by the chemical shift of H₂-1' ($\delta_{\rm H}$ 4.30) and its HMBC correlation (Fig. 2) to C-7 ($\delta_{\rm C}$ 166.5). Compound **2** was thus characterized to be (*Z*)-3-hexenyl syringic acid ester.

		1		2
No		1		2
	δc	δ_{H}	δc	δ _H
1	127.4		121.5	
2	112.9	7.76, d (1.9)	106.7	7.32, s
3	146.0		146.7	
4	147.1		139.3	
5	113.9	6.90, d (8.1)	146.7	
6	125.7	7.10, dd (8.1, 1.9)	106.7	7.32, s
7	143.7	6.79, d (12.8)	166.5	
8	117.1	5.82, d (12.8)		
9	166.8			
1'	64.6	4.12, t (6.8)	64.6	4.30, t (6.9)
2'	28.5	1.65, m	27.1	2.52, m
3'	28.3	1.33, m	123.9	5.40, dtt (10.7, 7.0, 1.5)
4′	22.5	1.33, m	134.8	5.54, dtt (10.7, 7.3, 1.5)
5'	14.1	0.90, t (6.8)	20.8	2.10, m
6′			14.4	0.97, t (7.5)
4-OH		5.83, brs		5.89, brs
3-OMe	56.1	3.93, s	56.6	3.94, s
5-OMe			56.6	3.94, s

Table 1. NMR Data of compounds 1 and 2 in CDCl₃

The four known compounds were identified to be methyl (*Z*)-ferulic acid ester (**3**) [24], *n*-butyl (*E*)-ferulic acid ester (**4**) [25], *n*-propyl (*E*)-ferulic acid ester (**5**) [25] and methyl (*E*)-ferulic acid ester (**6**) [25] by spectroscopic methods and comparison of NMR data with those of formerly reported compounds. All these aromatic constituents were screened for their antioxidant effect in the DPPH radical removing assay and NO release inhibitory activity in a lipopolysaccharide (LPS)-induced murine RAW264.7 macrophage model. While no tested compounds showed obvious NO production inhibitory activity, the two (*Z*)-ferulic acid esters **1** and **3** did exhibit significant antiradical effect with IC₅₀ of 25.0 ± 0.7 and $21.6 \pm 0.9 \mu$ M, respectively, being comparable to that (IC₅₀ = $25.4 \pm 2.9 \mu$ M) of the reference compound ascorbic acid.

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Supporting Information

Supporting information accompanies this paper on <u>http://www.acgpubs.org/journal/records-of-natural-products</u>

ORCID Xin-Xin Cao: 0000-0001-8362-4473 Jin-Yue Sun: 0000-0002-7377-0319 Chao Liu: <u>0000-0003-2680-1810</u> Jun-Sheng Zhang: <u>0000-0001-6626-1842</u> Hua Zhang: <u>0000-0002-5959-6473</u>

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