### **Supporting Information**

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# Impact of test conditions on the bacterial bioassay in the presence of TiO<sub>2</sub> nanoparticles

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Table S1: Chemical parameters analyzed in seawater samples, and related information about the procedure.

Parameter	Method	Instrument	Reference
SO4 <sup>2-</sup>	Turbidimetric as barium sulfate (375.4): Sulfate ion is converted to a barium sulfate suspension under controlled conditions. The resulting turbidity is determined spectrophotometrically at 420 nm.	UV-VIS spectrometry (Biochrom Libra S70 spectrophotometer)	[1]
NO3 <sup>-</sup>	Sulfanilamide/ethylenediamine with Cd reduction (353.3): The nitrite (that originally present plus reduced nitrate) is determined by diazotizing with sulfanilamide and coupling with N-(1-naphthyl)-ethylenediamine dihydrochloride to form a highly colored azo dye which is measured spectrophotometrically at 540 nm	UV-VIS spectrometry (Biochrom Libra S70 spectrophotometer)	[1]
$ m NH_4^+$	Nesselerization (APHA 4500): The sample is buffered at a pH of 9.5 with a borate in order to decrease hydrolysis of cyanates and organic nitrogen compounds and is then distilled into a solution of boric acid. The ammonia in the distillate is determined colorimetrically by Nesslerization at 425.0 nm by spectrometrically.	UV-VIS spectrometry (Biochrom Libra S70 spectrophotometer)	[1]
Cŀ	Chromatographic separations were performed at 30 °C with a Dionex IonPac AS20 analytical column $(2 \times 250 \text{ mm})$ . In addition, guard column and cartridge using ultra-pure (UP) water obtained from Dionex. The gradient programme: 10 mM of KOH for 6 min; linear increase of the KOH concentration from 10 mM to 25 mM for 15 min; 25 mM of KOH for 4 min; linear increase of the KOH concentration from 25 mM to 40 mM for 5 min; 40 mM of KOH for 5 min; linear decrease of the KOH concentration from 40 mM to 10 mM for 2 min. A 75 $\mu$ L-aliquot of the sample/standard solution was loaded into the eluent stream. Flow rate of 2.5 mL/min.	Ion chromatograpy (Dionex ICS-3000)	[2]
Na, K	Direct analysis of seawater samples according to the EPA 200.5	ICP-OES (Spectro, SpectroBlue)	[3]
PO <sub>4</sub> -	Ammonium molybdate solution acidified with $H_2SO_4$ was added onto the extracted samples along with excess ascorbic acid. The formation of the green/blue color was observed after heating them in the water bath. Colorimetric measurements were taken both at 822 nm and 650 nm for the purpose of comparison.	UV-VIS spectrometry (Biochrom Libra S70 spectrophotometer)	[4]

	Results	
Chemical Property		
	1% SW	100% SW
рН	8.0±0.9	8.3±0.8
Na (mg/L)	69±3	6693±535
K (mg/L)	ND	231±9
NO <sub>3</sub> (mg/L)	ND	0.26±0.02
NO <sub>2</sub> (mg/L)	ND	ND
NH <sub>3</sub> -N (mg/L)	ND	0.92±0.06
SO <sub>4</sub> (mg/L)	24±2	2283±190
PO <sub>4</sub> (mg/L)	ND	ND
Cl <sup>-</sup> (g/L)	ND	15.9±0.6

# **Table S2:** Chemical properties of seawater as a real environmental media (N:3, SW: seawater, ND: not detected).

#### References

- [1] APHA. American Public Health Association (2017). Standard methods for the examination of water and waste water. 23rd Edition, American Public Health Association, American Water Works Association, Water Environment Federation.
- [2] A. Baysal, H. Baltaci, N. Ozbek, O. Destanoglu, G. S. Ustabasi and G. Gumus (2017) Chemical characterization of surface snow in Istanbul (NW Turkey) and their association with atmospheric circulations, *Environ. Monit. Assess.* 189(275), 1-20.
- [3] EPA Method 200.5, Determination of trace elements in drinking water by axially viewed inductively coupled plasma-atomic emission spectrometry
- [4] EPA Method 365.3: Phosphorous, all forms (colorimetric, ascorbic acid, two reagent)