

Ent-Kaurane Diterpenoids from *Euphorbia hirta*

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Abstract: A new *ent*-kaurane diterpenoid was isolated from the ethanol extract of *Euphorbia hirta*, and elucidated as 2 β ,16 α ,19-trihydroxy-*ent*-kaurane (**1**), along with two known *ent*-kauranes, 2 β ,16 α -dihydroxy-*ent*-kaurane (**2**) and 16 α ,19-dihydroxy-*ent*-kaurane (**3**). Their structures were elucidated on the basis of spectroscopic analysis including 1D and 2D NMR techniques and HRESIMS evidence.

Keywords: Euphorbiaceae; *Euphorbia hirta*; *ent*-kaurane; diterpenoid.

1. Introduction

Euphorbia is the largest genus in the family of Euphorbiaceae, comprising about 2000 species. More than 80 of them are distributed in China [1]. Many secondary metabolites with specific types of diterpene skeletons in this genus have been found to possess a number of interesting biological activities [2-4]. *Euphorbia hirta* Linn, mainly distributed in the southern and southwestern district of China [5], has been used traditionally for the treatment of gastrointestinal disorders (diarrhea, dysentery, intestinal parasitosis), conjunctivitis, and respiratory diseases (asthma, bronchitis, hay fever) [6,7]. Previous phytochemical studies have reported six diterpenoids of abietane, ingenane and tigliane-types from the species [8,9]. On the course of our search for diterpenoids from *Euphorbia hirta*, a new *ent*-kaurane diterpenoid, 2 β ,16 α ,19-trihydroxy-*ent*-kaurane (**1**), as well as two known *ent*-kauranes (**2** and **3**) were obtained. Among them, compounds **2** and **3** were the first examples as natural products, and NMR data for **2** have not been reported so far. In this paper, we present the isolation and structural elucidation of the new compound (**1**) and the NMR data of **2**.

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2. Materials and Methods

2.1. General procedures

Optical rotations were measured using a Perkin-Elmer model 241 polarimeter. UV spectra were carried out on a Shimadzu UV-2401A spectrophotometer. IR spectra were recorded on a Bio-Rad FTS-135 spectrometer with KBr pellets. 1D and 2D NMR spectra were measured on a Bruker DRX-500 instrument with TMS as internal standard. Mass spectra were obtained on a VG Auto Spec-3000 spectrometer or on a Finnigan MAT 90 instrument. Column chromatography was performed on silica gel (200–300 mesh; Qingdao Marine Chemical Inc.), the Sephadex LH-20 (25–100 μm , Pharmacia Fine Chemical Co. Ltd.), and MCI-gel CHP 20P (75–150 μm , Mitsubishi Chemical Corp.). Thin-layer chromatography (TLC) was carried out on silica gel 60 F₂₅₄ on glass plates (Qingdao Marine Chemical Inc.) using various solvent systems.

2.2 Plant Material

The material of plant (*Euphorbia hirta*) was collected in the Xishuang Banna prefecture, Yunnan Province, People's Republic of China, in September 2009. A voucher specimen (YTCM 20090909) was deposited at the Yunnan Traditional Chinese Medical College, and was identified by Prof. Yao-wen Yang.

2.3 Extraction and Isolation

Dry herbs of *Euphorbia hirta* (4.0 kg) were extracted with ethanol (3 \times 25 L) at room temperature overnight. The extract was partitioned between H₂O and EtOAc, and an EtOAc layer (58 g) was chromatographed on MCI-gel CHP 20P (eluted with 90% CH₃OH–H₂O, then 100% CH₃OH). The 90% CH₃OH fraction (32 g) was repeatedly chromatographed over silica gel eluted in a step gradient manner with petroleum ether–acetone (1:0 to 0:1) to afford fractions A–F. Separation of fraction A (4 g) by silica gel column chromatography, eluted with petroleum ether–EtOAc (99:1→1:1), yielded fractions A1–A4. Compounds **2** (2 mg) and **3** (3 mg) were obtained from fraction A1 by repeated chromatography over silica gel with petroleum ether–EtOAc (49:1). Fraction B (5 g) was separated by silica gel column chromatography, eluted with petroleum ether–EtOAc (49:1→1:0), and yielded fractions B1–B4. Fraction B3 afforded compound **1** (2 mg) by chromatography over silica gel with petroleum ether–EtOAc (9:1) and the Sephadex LH-20 column chromatography (100% MeOH).

3. Results and Discussion

Compound **1** was obtained as white powder with $[\alpha]_{\text{D}}^{16.1}$ -60.79 ($c=0.16$, MeOH), and its molecular formula was determined to be C₂₀H₃₄O₃ by HRESIMS data (345.2413 [M + Na]⁺, calcd 345.2405). IR absorptions at 3386 cm⁻¹ indicated the presence of hydroxyl group and UV absorption at 202 nm revealed the absence of unsaturated moiety. The ¹³C NMR spectrum of **1** (Table 1) displayed 20 carbon signals corresponding to three methyls, nine methylenes, four methines and four quaternary carbons, which was assigned to be a diterpenoid skeleton. The diagnostic signals of three methines [δ_{C} 57.6 (C-5), 58.5 (C-9) and 49.0 (C-13)] and three quaternary carbons [δ_{C} 41.5 (C-4), 46.3 (C-8) and 41.9 (C-10)] indicated *ent*-kaurene skeleton of the diterpenoid. The COSY spectrum showed the presence of three proton sequences of -CH₂ (1)-CH (2)-CH₂ (3)-, -CH (5)-CH₂ (6)-CH₂ (7)- and -CH (9)-CH₂ (11)-CH₂ (12)-CH (13)-CH₂ (14)-, as shown in Figure 2. An oxygenated methylene (δ_{H} 3.62 and 3.32, each 1H, d, $J=11.2$ Hz; δ_{C} 65.7), an oxygenated methine (δ_{H} 3.87, 1H, m; δ_{C} 65.0) and an oxygenated quaternary carbon (δ_{C} 79.7) observed in the ¹³C NMR spectrum were located at C-19, C-2 and C-16, respectively, which was supported by the HMBC correlations from H-19 to C-4, C-5 and C-18, from H-2 to C-4 and C-10 and from H-12, H-14, H-15 and CH₃-17 to C-16 (Figure 2). In addition,

the shifts of C-2 were agreement with those of the known compound *ent*-2 α -hydroxy-8(14),15-pimaradiene [10]. These considerations helped in assigning the planar structure of **1** to be 2,16,19-trihydroxy-*ent*-kaurane.

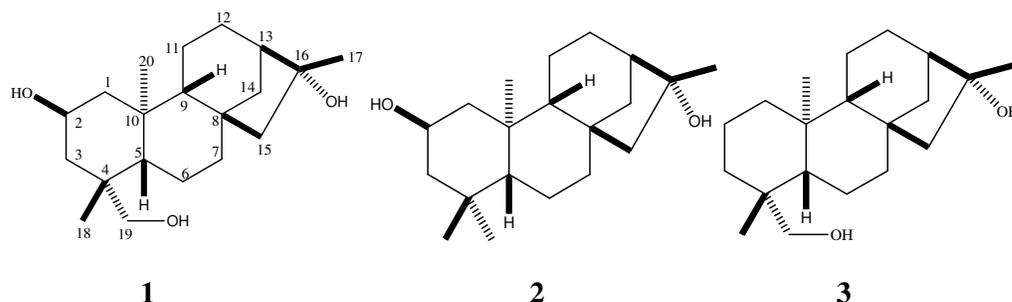


Figure 1. The structures of compounds **1-3**

The relative configuration of compound **1** was established on the basis of ROESY correlations of H-2 with H-19 and Me-20, H-19 with Me-20, Me-17 with H-12 β , H-12 β with H-9 β as well as 16-OH with H-13 α as shown in computer-generated 3D drawing (Figure 2), which suggested the 2-OH and 16-OH possess β and α -orientations, respectively. Moreover, the presence of 19-OH was proved by the downfield shift of C-5 (δ_C 57.6) owing to the absence of a γ -steric compression effect between 18 β -OH and H-5. Thus, compound **1** was unambiguously identified as 2 β ,16 α ,19-trihydroxy-*ent*-kaurane.

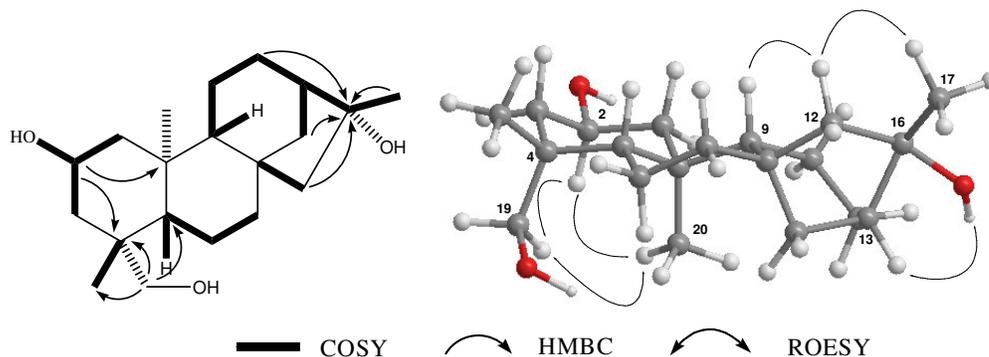


Figure 2. Key HMBC, COSY and ROESY correlations of compound **1**

Compounds **2** and **3** were identified according to the NMR and MS as well as comparison with values from the literature [11].

Table 1. $^1\text{H-NMR}$ and $^{13}\text{C-NMR}$ data of compounds **1** and **2** in CD_3OD ^a

1			2		
Position	$^1\text{H-NMR}$	$^{13}\text{C-NMR}$	Position	$^1\text{H-NMR}$	$^{13}\text{C-NMR}$
1 α	2.17, m	50.5 (CH ₂)	1 α	2.16, br d, $J=11.0$ Hz	52.0 (CH ₂)
1 β	0.68, t, $J=11.2$ Hz		1 β	0.65, t, $J=11.0$ Hz	
2 α	3.87, m	65.0 (CH)	2 α	3.86, m	65.5 (CH)
3 α	2.13, m	46.3 (CH ₂)	3 α	2.14, br d, $J=11.4$ Hz	50.4 (CH ₂)
3 β	0.82, t, $J=11.0$ Hz		3 β	1.83, br s	
4	-	41.5 (C)	4	-	35.7 (C)
5 β	0.93, d, $J=10.8$ Hz	57.6 (CH)	5 β	0.80, d, $J=11.8$ Hz	57.1 (CH)
6	1.68, m	21.3 (CH ₂)	6	1.46, m	21.2 (CH ₂)
7 α	1.64, overlap	43.5 (CH ₂)	7 α	1.62, overlap	43.1 (CH ₂)
7 β	1.45, m		7 β	1.47, m	
8	-	46.3 (C)	8	-	46.2 (C)
9 β	1.05, d, $J=4.8$ Hz	58.5 (CH)	9 β	1.05, d, $J=4.9$ Hz	58.3 (CH)
10	-	41.9 (C)	10	-	42.1 (C)
11	1.64, overlap	19.3 (CH ₂)	11	1.60, overlap	19.2 (CH ₂)
12	1.63, overlap	27.7 (CH ₂)	12	1.59, overlap	27.9 (CH ₂)
13 α	1.83, br s	49.0 (CH)	13 α	3.30, t, $J=1.4$ Hz	49.0 (CH)
14 α	1.87, d, $J=11.6$ Hz	38.4 (CH ₂)	14 α	1.89, d, $J=11.6$ Hz	38.6 (CH ₂)
14 β	1.60, overlap		14 β	1.58, overlap	
15	1.54, s	58.5 (CH ₂)	15	1.53, s	58.7 (CH ₂)
16	-	79.7 (C)	16	-	79.8 (C)
17	1.33, s	24.4 (CH ₃)	17	1.36, s	24.5 (CH ₃)
18	0.99, s	28.1 (CH ₃)	18	0.86, s	34.2 (CH ₃)
19 a	3.62, d, $J=11.2$ Hz	65.7 (CH ₂)	19	0.92, s	22.8 (CH ₃)
19 b	3.32, d, $J=11.2$ Hz		20	1.09, s	19.4 (CH ₃)
20	1.08, s	19.9 (CH ₃)			

^aThe assignments are based on DEPT, HSQC and HMBC experiments.

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Supporting Information:

The following Supporting Information is available for this article: <http://www.acgpubs.org/RNP>

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