

Flavonoids from Twigs of *Millettia leptobotrya* Dunn.

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Abstract: A new furanoisoflavone, 2'-methoxy-4',5'-methylenedioxy-[2",3":7,8] furanoisoflavone, leptobotryanone (**1**), and a new natural *O*-prenylated isoflavone, 4'- γ , γ -dimethylallyloxy-5,7-dihydroxyisoflavone (**2**), were isolated from the twigs of *Millettia leptobotrya*, together with twelve known flavonoids, 4'- γ , γ -dimethylallyloxy-5-hydroxy-7-methoxyisoflavone (**3**), 2',6,7-trimethoxy-4',5'-methylenedioxy-isoflavone (**4**), 2',7-dimethoxy-4',5'-methylenedioxyisoflavone (**5**), maximaisoflavone B (**6**), medicarpin (**7**), maackiain (**8**), genistein (**9**), biochanin A (**10**), prunetin (**11**), chrysoeriol (**12**), kaempferol (**13**) and desmoxyphyllin A (**14**). The structures of new compounds were elucidated on the basis of spectroscopic data interpretation, including 1D and 2D NMR and HREIMS. This is the first phytochemical investigation of this plant.

Keywords: *Millettia leptobotrya* Dunn; Leguminosae; leptobotryanone; isoflavone.

1. Introduction

About two hundred species of *Millettia* (Leguminosae /Fabaceae) were distributed in subtropical and tropical Africa, Asia and Australia, and many species were used as medicinal drugs, insecticide, or for stupefying fish in China [1]. Plants of the genus *Millettia* are well known for elaborating prenylated flavones and isoflavones with annellated furan and pyran rings [2]. *Millettia leptobotrya* Dunn is a tree distributed in south of Yunnan Province, China. The roots and leaves of this plant have been used by the local people for the treatment of fracture, traumatic injury and rheumatoid arthritis [3]. There have been no reports on the chemical composition of this plant so far. As part of our continuing studies on bioactive compounds from tropical medicinal plants, we have first examined the twigs of the title plant and isolated a new furanoisoflavone, leptobotryanone (**1**), and a new natural product, 4'- γ , γ -dimethylallyloxy-5,7-dihydroxyisoflavone (**2**), along with twelve known analogues. In the present paper, we report the isolation and structure elucidation of the new compounds.

2. Materials and Methods

2.1. General

TLC was performed with silica gel GF254 (Marine Chemical Industry Factory, Qingdao, China), and the spots were detected with a UV₂₅₄ lamp and by heating silica gel plates sprayed with 10% H₂SO₄ in ethanol. Column chromatography was performed using silica gel (200-300 mesh, Qingdao Haiyang Chemical Co., Ltd, Qingdao, China), reverse-phase C18 silica gel (40-63 μ m, Merck, Darmstadt Germany) and Sephadex LH-20 (GE healthcare, Sweden), MCI-gel CHP 20P (75–150 μ m;

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Mitsubishi Chemical Co. Japan). UV spectra were measured with a Shimadzu UV-2401 PC spectrophotometer. IR spectra were recorded on a Bruker Tensor-27 infrared spectrophotometer with KBr pellets. All NMR experiments were performed on a Bruker AM-400 and DRX-500 instruments with TMS as the internal standard. HREIMS spectra were recorded on a Waters AutoSpec Premier P776 instrument.

2.2. Plant Material

The twigs of *Millettia leptobotrya* were collected from Xishuangbanna, Yunnan Province, P.R. China in February 2012, and authenticated by Prof. Hong Wang, herbarium of Xishuangbanna Tropical Botanical Garden, Chinese Academy of Sciences. A voucher specimen (No 20120202) was deposited in the ethnobotany research group of Xishuangbanna Tropical Botanical Garden, Chinese Academy of Sciences.

2.3. Extraction and Isolation

The sun dried and powdered twigs of *M. leptobotrya* (6.5 kg) were extracted three times by maceration with 95% EtOH at room temperature, to afford crude extract after evaporation under vacuum. The crude extract was suspended in water and successively extracted with chloroform (CHCl₃) and ethyl acetate (EtOAc). The combined CHCl₃ extract was evaporated to give a deep-brown gum (148 g), which was separated on a silica gel column chromatography (CC) using petroleum ether (PE)-EtOAc step-gradient elution (9:1→ 4:6) to yield 5 fractions (C1, C2, C3, C4, C5). The fraction C1 (2.5 g 9:1) was further separated by reverse-phase C18 silica gel (RP-18) chromatography (eluted with 80–90% MeOH) and obtained the compound **3** (47 mg). The fraction C2 (18 g 8:2) was subjected to a silica gel CC again using PE - EtOAc (9:1, 85:15, 8:2, 6:4). The fraction (85:15) was applied to RP-18 CC (eluted with 70–90% MeOH/H₂O) and yielded the compounds **1** (17 mg) and **2** (22 mg). The part (8:2) was recrystallized to yield compounds **6** (32 mg). The part (6:4) was further purified by Sephadex LH-20 (MeOH) to afford compounds **7** (21 mg) and **8** (11 mg). The fraction C3 (15 g 7:3) was submitted to silica gel CC using PE-EtOAc (4:1–7:3–4:6) as elute. Compound **5** (25 mg) was obtained from the part (4:1) and the mixture of **10** and **11** (51 mg) was obtained from the part eluted (4:6) and further separated by RP-18 CC (eluted with 70–80% MeOH/H₂O). The fraction C4 (12 g 6:4) was subjected to silica gel CC eluted with CHCl₃-MeOH (25: 1) to yield **4** (19 mg) and **9** (24 mg). The fraction C5 (11 g 4:6) was applied to silica gel CC eluted with CHCl₃-MeOH (15: 1→10: 1) to afford **12** (19 mg), **13** (36 mg) and **14** (15 mg).

leptobotryanone (**1**): Pale yellow amorphous powder, UV (MeOH): λ_{\max} nm (log ϵ): 304 (3.92), 234 (4.37); IR (KBr): ν_{\max} : 1641, 1584, 1504, 1461, 1405, 1346, 1268, 1194 cm⁻¹. HREIMS: m/z 336.0629 (calc. for C₁₉H₁₂O₆, 336.0634). ¹H and ¹³C NMR see the Table 1.

4'- γ,γ -dimethylallyloxy-5,7-dihydroxyisoflavone (**2**): Colorless amorphous powder; UV (MeOH): λ_{\max} nm (log ϵ): 326 (sh) (3.49), 266 (4.40); IR (KBr): ν_{\max} : 3440, 2854, 1652, 1617, 1577, 1516, 1436, 1370, 1284, 1245, 1181 cm⁻¹. HREIMS: m/z 338.1161 (calc. for C₂₀H₁₈O₅, 338.1154). ¹H and ¹³C NMR see the Table 1.

3. Results and Discussion

Repeated column chromatography (including normal-phase silica gel, RP-18 silica gel and Sephadex LH-20) of the EtOH extract of the twigs of *M. leptobotrya* has led to the isolation of a new furanoisoflavone, leptobotryanone (**1**), and a new natural product, 4'- γ,γ -dimethylallyloxy-5,7-dihydroxyisoflavone (**2**), together with twelve known flavonoids, 4'- γ,γ -dimethylallyloxy-5-hydroxy-7-methoxyisoflavone (**3**) [4], 2',6,7-trimethoxy-4',5'-methylenedioxyisoflavone (**4**) [5], 2',7-dimethoxy-4',5'-methylenedioxyisoflavone (**5**) [6], maximaisoflavone B (**6**) [7], medicarpin (**7**) [8], maackiain (**8**) [9], genistein (**9**) [10], biochanin A (**10**) [11], prunetin (**11**) [12], chrysoeriol (**12**) [13], kaempferol (**13**) [13] and desmoxyphyllin A (**14**) [14]. The structures of the known compounds (**3-14**)

were characterized on the basis of spectral data and comparison with those reported in the literature. All compounds (Fig. 2) were isolated from *Millettia leptobotrya* for the first time, and compound **2** was the first example as natural product.

Compound **1** was obtained as pale yellow amorphous powder. The HREIMS exhibited a molecular ion peak at m/z 336.0629 (calcd. 336.0634), suggesting the molecular formula of $C_{19}H_{12}O_6$. The IR spectrum exhibited strong absorption band of a conjugated carbonyl at 1641 cm^{-1} . The UV spectrum showed strong absorptions at λ 304 and 234 nm. The ^{13}C NMR and DEPT spectrum revealed nineteen carbon signals corresponding to one methoxyl, one methylene, seven methine and ten quaternary carbons (including one carbonyl). The ^1H and ^{13}C NMR spectra exhibited characteristic signals at δ_{H} 8.04 (1H, *s*, H-2) and δ_{C} 153.7 (C-2), suggesting the existence of isoflavone skeleton [15] (Table 1). In addition, ^1H and ^{13}C NMR signals also revealed the presence of a furan ring at δ_{H} 7.14 (1H, *d*, $J = 1.2\text{ Hz}$, H-4'')/ δ_{C} 104.2 (C-4'') and δ_{H} 7.75 (1H, *d*, $J = 2.4\text{ Hz}$, H-5'')/ δ_{C} 145.6 (C-5'') as well as one methoxyl group at δ_{H} 3.74 (3H, *s*) / δ_{C} 56.8 and a methylenedioxy moiety at δ_{H} 5.97 (2H, *s*) / δ_{C} 101.4.

The appearance of H-5 and H-6 as doublets at δ_{H} 8.23 (1H, *d*, $J = 8.8\text{ Hz}$) and 7.56 (1H, *d*, $J = 8.8\text{ Hz}$) and the HMBC correlations of H-5 with C-4 (δ_{C} 176.2) and C-7 (δ_{C} 158.1) indicated that the furan ring should be fused in an angular position at C-7 (oxygenated) and C-8 [16], which was supported by the HMBC correlations of H-4'' and H-5'' with C-7 and H-6 and H-5'' with C-8 (δ_{C} 117.0). The relative positions of substituted groups on ring B were determined on the basis of the HMBC spectrum (Fig. 1). The methoxyl group was placed at C-2' as its proton (δ_{H} 3.74) showed 3J correlation with C-2' (δ_{C} 152.9) in the HMBC spectrum. The methylenedioxy unit should be located at C-4', C-5', which was confirmed by the HMBC 3J correlations of two protons (OCH₂O) at δ_{H} 5.97 with C-4' (δ_{C} 148.5) and 5' (δ_{C} 141.2). Based on the above spectral evidence, compound **1** was assigned as 2'-methoxy-4',5'-methylenedioxy-[2'',3'':7,8] furanoisoflavone and has been given the trivial name leptobotryanone.

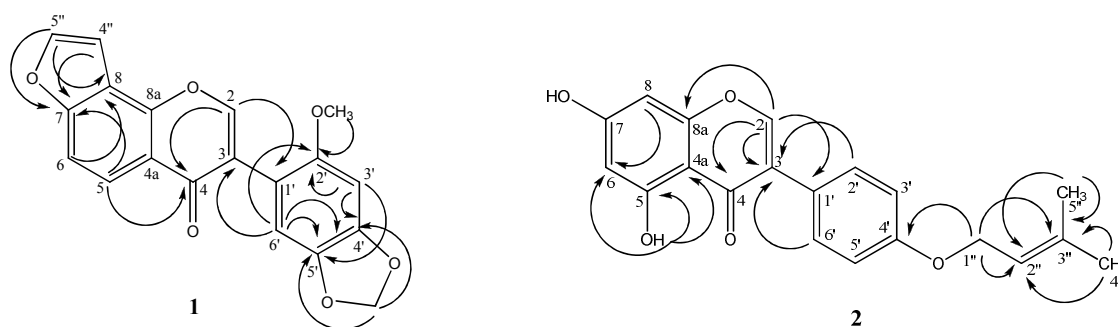


Figure 1. Key HMBC (H→C) correlations of compounds **1** and **2**.

Compound **2** was isolated as colorless amorphous powder. The molecular formula of **2** was determined to be $C_{20}H_{18}O_5$ by HREIMS: m/z 338.1161 [M]⁺, calcd. 338.1154. Its IR spectrum showed the presence of a chelated hydroxyl (3440 cm^{-1}) and a conjugated carbonyl (1652 cm^{-1}). The UV spectrum showed absorptions at λ_{max} 326 (sh) and 266 nm. The ^1H and ^{13}C NMR spectra of **2** (Table 1) exhibited low-field resonances at δ_{H} 8.19 (1H, *s*) and δ_{C} 154.4, which were characteristic of H-2 and C-2, respectively, of an isoflavone nucleus [15]. The HMBC spectrum (Fig. 1) showed the correlations of H-2 (δ_{H} 8.19) with C-3 (δ_{C} 124.0), C-4 (δ_{C} 181.5), C-8a (δ_{C} 159.0) and C-1' (δ_{C} 123.7), confirming its isoflavone structure.

In the ^1H NMR spectrum, a downfield signal at δ_{H} 13.01 was assigned to OH-5 by the HMBC correlations of δ_{H} 13.01 with C-4a (δ_{C} 106.0), C-5 (δ_{C} 163.8) and C-6 (δ_{C} 99.8); two *meta*-coupled aromatic protons at δ_{H} 6.28 and 6.41 (each 1H, *d*, $J = 1.6\text{ Hz}$) were attributed to H-6 and H-8 in ring A, respectively. Moreover, two sets of *ortho*-coupled doublets ($J = 8.4$ and 8.8 Hz) at δ_{H} 6.98 and 7.52, integrating for two protons each, were attributed to H-3', H-5' and H-2', H-6', respectively, indicating the presence of a *para*-disubstituted ring B in compound **2**. The signals at δ_{H} 5.47 (1H, *t*, $J = 6.7\text{ Hz}$), 4.59 (2H, *d*, $J = 6.4\text{ Hz}$), 1.77 (3H, *s*) and 1.75 (3H, *s*), assigned to methine, methylene and *gem*-

dimethyl protons, respectively, revealed the presence of an oxyprenyl residue in **2** [17]. The oxyprenyl moiety was connected to C-4' as the oxygenated methylene protons at δ_{H} 4.59 (CH₂-1'') correlated to C-4' (δ_{C} 159.8) of the isoflavone nucleus in the HMBC spectrum. Beside the partial structures mentioned above, the molecular formula C₂₀H₁₈O₅ requires a hydroxyl group which should be connected with C-7, which was a quaternary carbon and chemical shift appeared at δ_{C} 165.1 ppm in the low magnetic field. Therefore, compound **2** was identified as 4'- γ , γ -dimethylallyloxy-5,7-dihydroxyisoflavone.

Although compound **2** has been reported synthetically [18], this is the first time that it has been found as a naturally occurring compound and fully characterized. Prior to this study, the ¹H and ¹³C NMR data of **2** have not been reported.

Table 1. ¹H (400 MHz) and ¹³C (100 MHz) NMR data of **1** (in CDCl₃) and **2** in (Me₂CO-*d*₆).

Position	1		2	
	δ_{H} (J in Hz)	δ_{C}	δ_{H} (J in Hz)	δ_{C}
2	8.04 (1H, <i>s</i>)	153.7	8.19 (1H, <i>s</i>)	154.4
3		122.7		124.0
4		176.2		181.5
4a		119.9		106.0
5	8.23 (1H, <i>d</i> , <i>J</i> = 8.8)	122.5		163.8
6	7.56 (1H, <i>d</i> , <i>J</i> = 8.8)	110.1	6.28 (1H, <i>d</i> , <i>J</i> = 1.6)	99.8
7		158.1		165.1
8		117.0	6.41 (1H, <i>d</i> , <i>J</i> = 1.6)	94.4
8a		156.3		159.0
1'		112.6		123.7
2'		152.9	7.52 (1H, <i>d</i> , <i>J</i> = 8.8)	131.0
3'	6.64 (1H, <i>s</i>)	95.4	6.98 (1H, <i>d</i> , <i>J</i> = 8.4)	115.2
4'		148.5		159.8
5'		141.2	6.98 (1H, <i>d</i> , <i>J</i> = 8.4)	115.2
6'	6.86 (1H, <i>s</i>)	111.2	7.52 (1H, <i>d</i> , <i>J</i> = 8.8)	131.0
1''			4.59 (2H, <i>d</i> , <i>J</i> = 6.4)	65.3
2''			5.47 (1H, <i>t</i> , <i>J</i> = 6.7)	121.0
3''				137.8
4''	7.14 (<i>d</i> , <i>J</i> = 1.2)	104.2	1.75 (3H, <i>s</i>)	25.8
5''	7.75 (<i>d</i> , <i>J</i> = 2.4)	145.6	1.77 (3H, <i>s</i>)	18.2
5-OH			13.01 (1H, <i>s</i>)	
2'-OMe	3.74 (3H, <i>s</i>)	56.8		
OCH ₂ O	5.97 (2H, <i>s</i>)	101.4		

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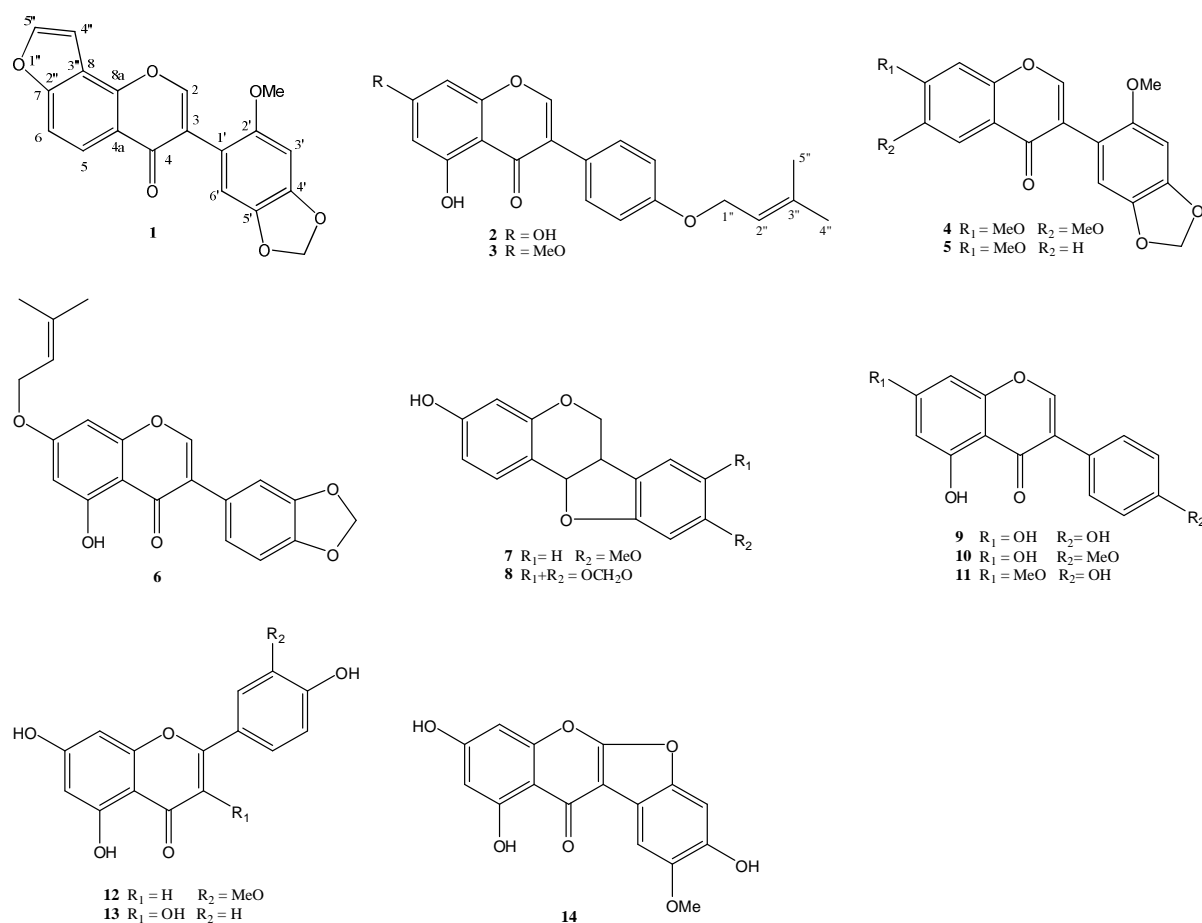


Figure 2. Structures of compounds 1-14.

Acknowledgments

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Supporting Information

Supporting Information accompanies this paper on <http://www.acgpubs.org/RNP>

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