

## Three New Polyketides from the Insect-Associated Fungus *Letendraea* sp. 5XNZ4-2

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(Received September 01, 2020; Revised October 15, 2020; Accepted October 16, 2020)

**Abstract:** Chemical investigation of the EtOAc extract of an insect-associated fungus *Letendraea* sp. 5XNZ4-2 cultured in Potato Dextrose Broth (1/2 PDB) medium lead to the isolation of three new polyketides, named letendronol D (**1**), phomopsiketones H-I (**2-3**). The structures of new compounds were elucidated by the analysis of HRESIMS and NMR spectroscopic data, and the absolute configurations were determined by modified Mosher's method, ECD calculation and single-crystal X-ray diffraction. Cytotoxicity and antibacterial activities of **1** were assayed and regrettably **1** didn't display any cytotoxicity and antibacterial activity. **3** was the first phomopsiketone derivative obtaining the lactone.

**Keywords:** polyketides; insect-associated fungus; *Letendraea* sp. © 2020 ACG Publications. All rights reserved.

### 1. Introduction

Insect-associated fungi, which develop symbiotic relationships with their hosts [1], can provide biologically active and structurally interesting natural products [2], such as macrodiolides [3, 4], alkaloids [5], polyketides [6] and so on that likely protect insect hosts from infestation [7].

During the course of our efforts toward searching for structurally new and bioactive secondary metabolites from insect-associated fungi [8], nine polyketides have been isolated from the 1/2 PDB culture broth of endophytic fungus *Letendraea* sp. 5XNZ4-2 [9, 10], indicating that its metabolic pathway was unique. More studies were carried out for this strain to explore its metabolic potential. As a result, three new polyketides, letendronol D (**1**), phomopsiketone H (**2**) and phomopsiketone I (**3**), were isolated. Herein, we describe the isolation, structure identification, and bioactivity evaluation of the new compounds.

### 2. Materials and Methods

#### 2.1. Materials and Instruments [9]

Optical rotations were recorded on Rudolph research analytical AUTOPOL I. The ultraviolet and Electronic circular dichroism (ECD) spectra were measured on Shimadzu UV-1800 spectrophotometer and JASCO J-1500 circular dichroism, respectively. The infrared (IR) spectra were

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The article was published by ACG Publications

<http://www.acgpubs.org/journal/records-of-natural-products> May-June 2021 EISSN:1307-6167

DOI: <http://doi.org/10.25135/rnp.206.20.08.1786>

acquired from a Thermo Nicolet iS10. 1D and 2D NMR spectra were recorded on Bruker AVIII 500 MHz and JEOL 600Hz, both using TMS as the internal standard. HR-ESI-MS data were obtained from an Agilent 6224 TOF LC-MS. Analytical and preparative liquid chromatography were performed on Agilent 1260 and Agilent Technologies ProStar system, while C18 (Cosmosil, 5  $\mu$ m, 4.6  $\times$  250 mm) packing column was used for HPLC analysis. The column chromatography (CC) was performed on Silica gel (200–300 mesh, Qing Dao Hai Yang Chemical Group Co.).

The *Letendraea* sp. was isolated from the gut of a crab found on Zhairuoshan Island (N20.2920, E122.5), Zhejiang Province, China. The fungus was determined as *Letendraea* sp. by 26s rDNA sequence analysis (GenBank accession no. MK743951).

## 2.2. Fermentation and Isolation

The strain was static cultured in 500 mL Erlenmeyer-flasks each containing 200 mL of 1/2 PDB media (100 g potato extraction; 17 g artificial sea salt and 10 g dextrose of 1 L pure water) at 28 °C for 30 days. The total culture broth was 20 L.

The total culture broth (20 L) was filtered and extracted with an equal volume of EtOAc for 3 times to obtain 2.79 g metabolites extract. The extract was fractionated by silica gel column chromatography (CC) eluted in a gradient petroleum ether-EtOAc (20:1-1:1) to yield 10 fractions (Fr.1-10) based on TLC analysis. Fr.7 was further separated via preparative HPLC eluting with MeOH/H<sub>2</sub>O (40/60, v/v) at 8 mL/min to obtain four sub-fractions Fr.7.1-7.4. Sub-fraction Fr.7.3 (7 mg) was purified with semi-preparative HPLC (MeOH/H<sub>2</sub>O 30:70, 4 mL/min) and yielded **2** (2.55 mg). Fr.8 was initially separated by CC over silica gel with CH<sub>2</sub>Cl<sub>2</sub>-MeOH gradient from 80:1-5:1 based on TLC analysis to afford 10 sub-fractions Fr.8.1-8.10. Sub-fraction Fr.8.8 was purified by semi-preparative HPLC at 4 mL/min using MeOH/H<sub>2</sub>O (25/75, v/v) as the eluting solvents and got Fr.8.8.4 (23 mg). Sub-fraction Fr.8.8.4 was further purified by semi-preparative HPLC at 4 mL/min using CH<sub>3</sub>CN/H<sub>2</sub>O (15/85, v/v) as the eluting solvents and yielded compound **3** (4.6 mg). Fr.10 was purified by CC over silica gel using a gradient of CH<sub>2</sub>Cl<sub>2</sub>-MeOH (50:1-1:1) as a mobile phase to provide five fractions (Fr.11 to 15). Fr.13 was separated via preparative HPLC eluting with MeOH/H<sub>2</sub>O (15:85, v/v) at 10 mL/min to obtain **1** (148 mg).

## 2.3. Spectral Data

*Letendronol D* (**1**): White amorphous powder; molecular formula C<sub>12</sub>H<sub>20</sub>O<sub>4</sub>; [ $\alpha$ ]<sub>D</sub><sup>20</sup> -6 (c 0.1, MeOH); ECD (0.50 mg/mL, MeOH)  $\lambda_{\max}$  ( $\Delta \epsilon$ ) 209 (-53.97) nm; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 259 (2.98) nm; IR ( $\lambda_{\max}$ ) 3316, 2954, 2935, 2864, 1648, 1450, 1418, 1379, 1341, 1275, 1236, 1186, 1119, 1030, 942, 889, 839 cm<sup>-1</sup>; <sup>1</sup>H NMR data (500 MHz, in CD<sub>3</sub>OD) and <sup>13</sup>C NMR data (125 MHz, in CD<sub>3</sub>OD), see Table 1; HRESIMS  $m/z$  [M-H]<sup>-</sup> 227.1292 (calcd for C<sub>12</sub>H<sub>19</sub>O<sub>4</sub>, 227.1283).

*Phomopsiketone H* (**2**): Colorless crystal in methanol; mp 116-116.5 °C; molecular formula C<sub>12</sub>H<sub>18</sub>O<sub>4</sub>; [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 1.83 (c 0.5, MeOH); ECD (0.50 mg/mL, MeOH)  $\lambda_{\max}$  ( $\Delta \epsilon$ ) 399 (-0.35), 338 (+4.41), 256 (-53.48), 224 (+50.72) nm; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 234 (3.95) nm; IR ( $\lambda_{\max}$ ) 3329, 2947, 2835, 1661, 1450, 1398, 1107 cm<sup>-1</sup>; <sup>1</sup>H NMR data (500 MHz, in CD<sub>3</sub>OD) and <sup>13</sup>C NMR data (125 MHz, in CD<sub>3</sub>OD), see Table 1; HRESIMS  $m/z$  [M+Na]<sup>+</sup> 249.1100 (calcd for C<sub>12</sub>H<sub>18</sub>O<sub>4</sub>Na, 249.1103).

*Phomopsiketone I* (**3**): White amorphous powder; molecular formula C<sub>12</sub>H<sub>18</sub>O<sub>4</sub>; [ $\alpha$ ]<sub>D</sub><sup>20</sup> + 66.18 (c 0.5, MeOH); ECD (0.50 mg/mL, MeOH)  $\lambda_{\max}$  ( $\Delta \epsilon$ ) 399 (-0.35), 338 (+4.41), 256 (-53.48), 224 (+50.72) nm; UV (MeOH)  $\lambda_{\max}$  (log  $\epsilon$ ) 215 (3.86) nm; IR ( $\lambda_{\max}$ ) 3334, 2960, 1646, 1403, 1260, 1209, 1170, 1089, 973 cm<sup>-1</sup>; <sup>1</sup>H NMR data (600 MHz, in CD<sub>3</sub>OD) and <sup>13</sup>C NMR data (150 MHz, in CD<sub>3</sub>OD), see Table 1; HRESIMS  $m/z$  [M+Na]<sup>+</sup> 249.1103 (calcd for C<sub>12</sub>H<sub>18</sub>O<sub>4</sub>Na, 249.1103).

## 2.4. Preparation of MTPA esters of Compounds **1**

Two parts of compound **1** (4.5 mg) were dissolved with 0.5 mL anhydrous pyridine and then react with (*R*)- or (*S*)-MTPA chloride (50  $\mu$ L), respectively. Each reaction mixture was stirred at

ambient temperature for 4 h and was terminated by adding 1 mL methanol. HPLC was also used for the isolation of 4, 7, 10-*tri-S*-MTPA ester and 4, 7, 10-*tri-R*-MTPA ester of **1**.

### 2.5. ECD Calculation of **3**

Conformational analyses were carried out via random searching in the Sybyl-X 2.0 using the MMFF94S force field with an energy cutoff of 2.0 kcal/mol [11]. The results showed 2 (C1, C2) lowest energy conformers for 4*R*, 7*S*, 10*R*-**3** and 2 (C3, C4) for 4*R*, 7*S*, 10*S*-**3**. Subsequently, the conformers were reoptimized using DFT at b3lyp/6-31+g (d,p) level in MeOH by the GAUSSIAN 09 program. The energies, oscillator strengths, and rotational strengths (velocity) of the first 30 electronic excitations were calculated using the TDDFT methodology at the *cam-b3lyp/TZVP* level using the polarizable continuum model in MeOH. The ECD spectrum were simulated by the overlapping Gaussian function (half the bandwidth at 1/e peak height,  $\sigma = 0.2$ ). To get the final spectra, the simulated spectra of the conformers were averaged according to the Boltzmann distribution theory and their relative Gibbs free energy ( $\Delta G$ ). Theoretical ECD spectra of the corresponding enantiomers (4*S*, 7*R*, 10*S*-**3** and 4*S*, 7*R*, 10*R*-**3**) were obtained by directly inverse of the ECD spectrum of 4*R*, 7*S*, 10*R*-**3** and 4*R*, 7*S*, 10*S*-**3**, respectively.

### 2.6. X-ray Crystallographic Analysis of **2**

Compound **2** was obtained as colorless crystals from methanol. X-ray single-crystal diffraction data of **2** was selected on a Bruker APEX-II CCD diffractometer at 170 K. Using Olex2 [12], the structure was solved with the ShelXT [13] structure solution program using Intrinsic Phasing and refined with the ShelXL [14] refinement package using Least Squares minimisation. Crystallographic data for **2** has been deposited in the Cambridge Crystallographic Data Centre database (CCDC Number: 2008387).

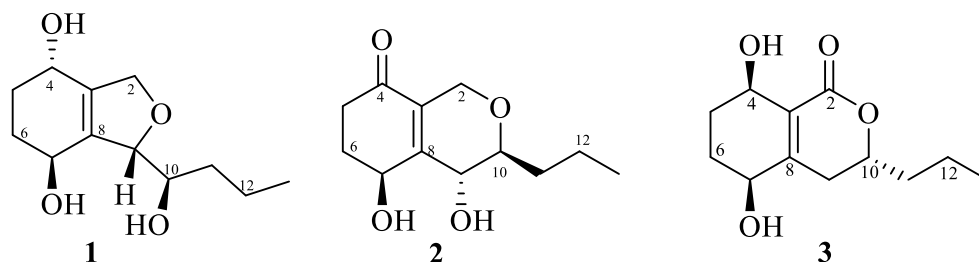
*Crystal Data of 2*: C<sub>12</sub>H<sub>18</sub>O<sub>4</sub> (*M* = 226.26 g/mol): monoclinic, space group C2 (no. 5), *a* = 20.2305 (12) Å, *b* = 7.6248 (5) Å, *c* = 8.3334 (5) Å,  $\beta$  = 113.6120(10)°, *V* = 1177.84 (13) Å<sup>3</sup>, *Z* = 4, *T* = 170.0 K,  $\mu$  (CuK $\alpha$ ) = 0.783 mm<sup>-1</sup>, *D*<sub>calc</sub> = 1.276 g/cm<sup>3</sup>, 9022 reflections measured (19.152° ≤ 2 $\theta$  ≤ 136.74°), 2097 unique (*R*<sub>int</sub> = 0.0175, *R*<sub>sigma</sub> = 0.0151) which were used in all calculations. *F* (000) = 488.0. The final *R*<sub>1</sub> was 0.0281 (*I* > 2 $\sigma$ (*I*)) and *wR*<sub>2</sub> was 0.0800 (all data). Flack parameter = 0.13 (3).

## 3. Results and Discussion

### 3.1. Structure Elucidation

Compound **1** was obtained as a white amorphous powder, and has a molecular formula of C<sub>12</sub>H<sub>20</sub>O<sub>4</sub> (with 3 degrees of unsaturation) deduced from its HRESIMS (*m/z* 227.1292 for [M-H]<sup>-</sup>) and NMR data. <sup>1</sup>H NMR (Table 1) of **1** displayed one methyl ( $\delta_{\text{H}}$  0.95, t, *J* = 7.0 Hz). The analysis of <sup>13</sup>C NMR and DEPT revealed 12 carbon signals, including two olefinic carbons ( $\delta_{\text{C}}$  138.4, 139.9), four oxygenated methine ( $\delta_{\text{C}}$  65.1, 65.5, 74.3, 90.7), one oxygenated methylene ( $\delta_{\text{C}}$  76.2), four methylene ( $\delta_{\text{C}}$  20.1, 31.7, 32.1, 34.8) and one methyl ( $\delta_{\text{C}}$  14.4). These signals were similar to those of letendronol A [9]. The same cyclohexene moiety was derived from the <sup>1</sup>H-<sup>1</sup>H COSY correlations between H-4 ( $\delta_{\text{H}}$  4.27)/H<sub>2</sub>-5 ( $\delta_{\text{H}}$  1.57, 2.09)/H<sub>2</sub>-6 ( $\delta_{\text{H}}$  1.55, 2.11)/H-7 ( $\delta_{\text{H}}$  4.30), coupled with the HMBC correlations from H<sub>2</sub>-5 to C-3 ( $\delta_{\text{C}}$  139.9) and H<sub>2</sub>-6 to C-8 ( $\delta_{\text{C}}$  138.4) (Figure 2). The similar CH<sub>3</sub>(13)-CH<sub>2</sub>(12)-CH<sub>2</sub>(11)-CHO(10)-CHO(9)-aliphatic chain, derived from <sup>1</sup>H-<sup>1</sup>H COSY correlations of H<sub>3</sub>-13 ( $\delta_{\text{H}}$  0.95)/H<sub>2</sub>-12 ( $\delta_{\text{H}}$  1.38, 1.60)/H<sub>2</sub>-11 ( $\delta_{\text{H}}$  1.48, 1.51)/H-10 ( $\delta_{\text{H}}$  3.71)/H-9 ( $\delta_{\text{H}}$  4.91), was positioned at C-8 according to the HMBC correlation from H-10 to C-8. Meanwhile, C-2 was connected with C-3 because of the HMBC correlations from H<sub>2</sub>-2 ( $\delta_{\text{H}}$  4.55, 4.77) to C-8, C-3. A dihydrofuran ring was formed by the HMBC correlation from H-9 to C-2, which was different from the dihydropyran ring in

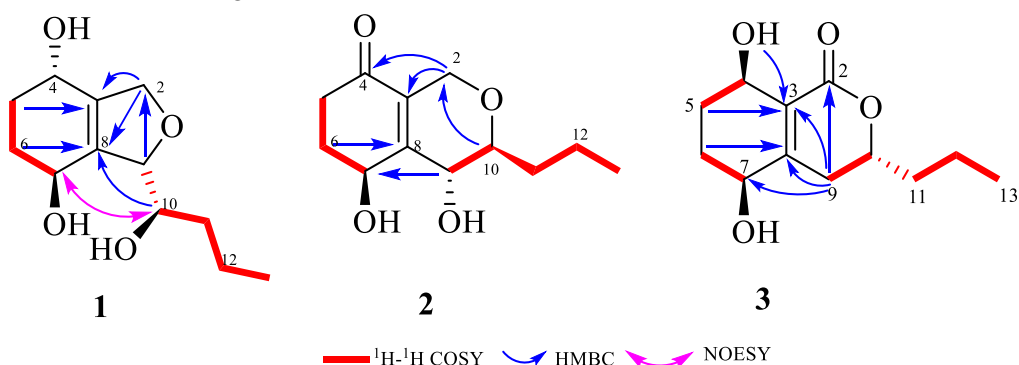
letendronol A. Thus, compound **1** was determined as a new polyketone and named as letendronol D (Figure 1).



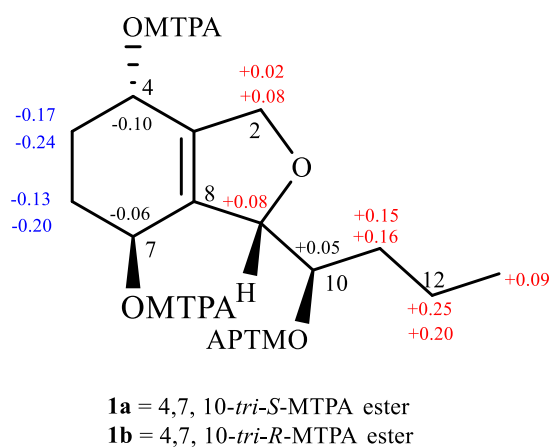
**Figure 1.** Chemical structures of compounds **1-3**

In the NOESY experiment of **1**, the correlation between H-7/H-10 suggested that H-7 and H-9 adopted different orientations with each other.

The absolute configuration of **1** was determined by a modified Mosher's esterification method [15] and esters were purified with preparative HPLC. The adducts were determined as 4,7,10-*tri-S*-MTPA ester (**1a**), 4,7,10-*tri-R*-MTPA ester (**1b**), respectively, by HRESIMS (4,7,10-*tri-S*-MTPA ester  $m/z$  899.2454; 4,7,10-*tri-R*-MTPA ester  $m/z$  899.2445 for  $[M+Na]^+$ , Figures S39 and S40). The  $\Delta\delta$  values ( $\Delta\delta_{1a-1b}$ , Figure 3) between the MTPA adducts (**1a/1b**) showed noticeable differentiation around C-4 (negative values for H<sub>2</sub>-5 and positive values for H<sub>2</sub>-2), C-7 (negative values for H<sub>2</sub>-6 and positive values for H-9) and C-10 (positive values for H<sub>2</sub>-11, H<sub>2</sub>-12 and H<sub>3</sub>-13), confirming the 4*S*, 7*S*, and 10*R* configurations. The NOESY correlation between H-7 and H-10 deduced the configuration of C-9 as *S*. Thus, the absolute configuration of **1** was determined as (4*S*, 7*S*, 9*S*, 10*R*).



**Figure 2.** <sup>1</sup>H-<sup>1</sup>H COSY, key HMBC and NOESY correlations of **1-3**



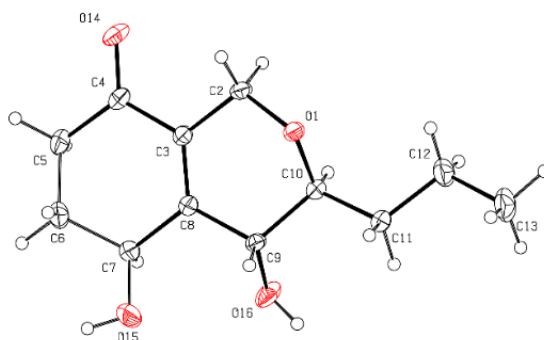
**Figure 3.**  $\Delta\delta_{S-R}$  values for the MTPA esters (**1a/1b**)

**Table 1.** NMR data of compounds **1-3**

Position	1 <sup>a</sup> (in CD <sub>3</sub> OD)		2 <sup>b</sup> (in CD <sub>3</sub> OD)		3 <sup>b</sup> (in DMSO)	
	$\delta_C$ , type	$\delta_H$ , m (J in Hz)	$\delta_C$ , type	$\delta_H$ , m (J in Hz)	$\delta_C$ , type	$\delta_H$ , m (J in Hz)
2	76.2 CH <sub>2</sub>	4.55, m 4.77, m	64.2 CH <sub>2</sub>	4.11, dt (16.4, 2.6) 4.34, dt (16.4, 2.6)	166.0 C	
3	139.9 C		133.4 C		126.1 C	
4	65.1 CH	4.27, m	199.5 C		60.9 CH	4.23, br s
5	31.7 CH <sub>2</sub>	1.57, m 2.09, m	35.1 CH <sub>2</sub>	2.35, dq (16.0, 4.6) 2.58, ddd (16.0, 7.4, 4.6)	29.7 CH <sub>2</sub>	1.52, m 1.70, m
6	32.1 CH <sub>2</sub>	1.55, m 2.11, m	32.2 CH <sub>2</sub>	1.99, m 2.24, m	26.8 CH <sub>2</sub>	1.74, m 1.79, m
7	65.5 CH	4.30, m	64.4 CH	4.71, m	68.7 CH	4.04, m
8	138.4 C		158.2 C		157.6 C	
9	90.7 CH	4.91, m	79.6 CH	3.24, td (8.2, 2.7)	30.5 CH <sub>2</sub>	2.38, overlapped
10	74.3 CH	3.71, dt (8.6, 4.0)	67.4 CH	4.17, d (8.1)	77.2 CH	4.33, m
11	34.8 CH <sub>2</sub>	1.48, m 1.51, m	35.5 CH <sub>2</sub>	1.47, m 1.81, m	37.2 CH <sub>2</sub>	1.56, m 1.65, m
12	20.1 CH <sub>2</sub>	1.38, m 1.60, m	19.8 CH <sub>2</sub>	1.42, m 1.58, m	18.7 CH <sub>2</sub>	1.35, m 1.41, m
13	14.4 CH <sub>3</sub>	0.95, t (7.0)	14.4 CH <sub>3</sub>	0.96, t (7.2)	14.6 CH <sub>3</sub>	0.91, t (7.4)
C4-OH						4.60, d (4.3)
C7-OH						5.23, d (5.6)

<sup>a</sup>Measured at 500 MHz NMR. <sup>b</sup>Measured at 600 MHz NMR.

Compound **2** was obtained as a colorless crystal in methanol. The molecular formula of **2** was determined as C<sub>12</sub>H<sub>18</sub>O<sub>4</sub>, according to its HRESIMS ( $m/z$  249.1100 for [M+Na]<sup>+</sup>). <sup>13</sup>C NMR (Table 1) of **2** displayed 12 carbon signals, including two olefinic carbons ( $\delta_C$  158.2, 133.4), one oxygenated methylene ( $\delta_C$  64.2), three oxygenated methine ( $\delta_C$  64.4, 67.4, 79.6), four methylene ( $\delta_C$  19.8, 32.2, 35.1, 35.5) and one methyl ( $\delta_C$  14.4), which were similar to those of phomopsiketone D [9]. **2** also had the same C<sub>5</sub> aliphatic chain and cyclohexene moiety according to the 2D NMR (Figure 2). While a dihydropyran ring was formed by the HMBC correlation from H-10 ( $\delta_H$  4.17) to C-2 ( $\delta_C$  64.2), which was different from dihydrofuran in phomopsiketone D. Thus, **2** was also a new family member of phomopsiketones and named as phomopsiketone H (Figure 1).

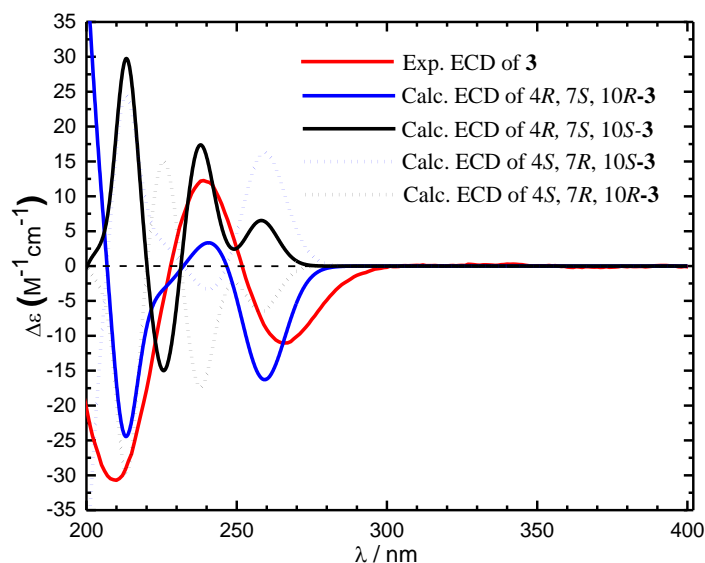
**Figure 4.** X-ray crystal structure of **2** (Flack parameter = 0.13(3))

The vicinal coupling constant  $J_{\text{H-9/H-10}}$  (8.1 Hz) indicated the *trans* relationship between H-9 and H-10 [16]. The configuration of **2** was unambiguously confirmed as (7*S*, 9*R* and 10*S*) by X-ray analysis (Figure 4).

Compound **3** was contained as white amorphous powder and has the same molecular formula of  $\text{C}_{12}\text{H}_{18}\text{O}_4$  (with 4 degrees of unsaturation) as that of **2** according to its HRESIMS ( $m/z$  249.1103 for  $[\text{M}+\text{Na}]^+$ ) and NMR data. The analysis of  $^{13}\text{C}$  NMR and HSQC revealed two olefinic carbons ( $\delta_{\text{C}}$  126.1, 157.6) and one ester ( $\delta_{\text{C}}$  166.0). Calculation of unsaturation revealed that compound **3** also contained bicyclic skeleton. Comparison of 1D NMR data between **3** and **2** revealed that the ketone carbonyl and oxygenated methylene in **2** was replaced by oxygenated methine ( $\delta_{\text{C}}$  60.9, C-4 in **3**) and lactone ( $\delta_{\text{C}}$  166.0, C-2 in **3**) (Table 1), which was confirmed by  $^1\text{H}$ - $^1\text{H}$  COSY correlations of  $\text{H}_2$ -5 ( $\delta_{\text{H}}$  1.52, 1.70)/H-4 ( $\delta_{\text{H}}$  4.23) (Figure 2) as well as the HMBC correlation from  $\text{H}_2$ -9 ( $\delta_{\text{H}}$  2.38) to C-2. The similar  $\text{C}_5$  side chain as those in **1-2** was derived from  $^1\text{H}$ - $^1\text{H}$  COSY correlations between  $\text{H}_3$ -13 ( $\delta_{\text{H}}$  0.91)/ $\text{H}_2$ -12 ( $\delta_{\text{H}}$  1.35, 1.41)/ $\text{H}_2$ -11 ( $\delta_{\text{H}}$  1.56, 1.65)/H-10 ( $\delta_{\text{H}}$  4.33)/ $\text{H}_2$ -9 ( $\delta_{\text{H}}$  2.38) and connected at C-8 ( $\delta_{\text{C}}$  157.6) according to the HMBC correlations from H-9 to C-8, C-3 ( $\delta_{\text{C}}$  126.1) and C-7 ( $\delta_{\text{C}}$  68.7). Different with **1** and **2**, **3** was a new lactone and named as phomopsiketone I (Figure 1). **3** was the first phomopsiketone derivative obtaining the lactone.

The NOESY correlation between C4-OH and C7-OH suggested that H-4 and H-7 adopted same orientations (Figure S33).

The absolute configuration of **3** was established by the comparison between experimental ECD spectrum and the theoretically calculated values of four possible stereoisomers (4*R*, 7*S*, 10*R*)-**3**, (4*R*, 7*S*, 10*S*)-**3**, (4*S*, 7*R*, 10*R*)-**3** and (4*S*, 7*R*, 10*S*)-**3**. The experimental ECD (Figure 5) of **3** showed a negative Cotton effect at 265 nm, a positive Cotton effect at 240 nm and a negative Cotton effect at 210 nm, which matched well with the calculated value of (4*R*, 7*S*, 10*R*)-**3**, and contributed to determine the absolute configuration of **3** as (4*R*, 7*S*, 10*R*)



**Figure 5.** Comparison between calculated ECD spectra and experimental curves of **3**

## Acknowledgments

This work was supported by Natural Science Foundation of China (NSFC No.41406141).

## Supporting Information

Supporting information accompanies this paper on <http://www.acgpubs.org/journal/records-of-natural-products>

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