

Constituents of the Flower of *Maxillaria tenuifolia* and Their Anti-Diabetic Activity

Chia-Ying Li *

Department of Applied Chemistry, National Pingtung University, Pingtung City, Pingtung County
900393, Taiwan

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Abstract: The EtOAc extract of the flower of *Maxillaria tenuifolia*, which had shown potent α -glucosidase inhibitory, was subjected to activity-guided isolation to yield four compounds, 3,4-dihydroxy benzoic acid methyl ester (**1**), flavanthridin (**2**), vanillic acid (**3**) and mangiferin (**4**). Their structures were ascertained by comparison of their physical and spectroscopic properties with those reported in the literature. Among them, flavanthridin (**2**), vanillic acid (**3**) and mangiferin (**4**) showed significant α -glucosidase inhibitory activity compared to acarbose at the concentration 1.0 mg/mL. Besides, 3,4-dihydroxy benzoic acid methyl ester (**1**) and mangiferin (**4**) also showed more effective in against the peroxidation of linoleic acid in an aqueous system than trolox. These results indicated that the EtOAc extract of *M. tenuifolia* and the related constituents might have anti-diabetic effect by suppressing carbohydrate disintegration and could prevent damage to organisms by oxidative stress.

Keywords: α -Glucosidase; diabetes; *Maxillaria tenuifolia*; antioxidant. © 2021 ACG Publications. All rights reserved.

1. Plant Source

Maxillaria tenuifolia Lindl. (Orchidaceae) were obtained from Lan Hui Biotech Co. (Taiwan) and verified by Dr. Y. Y. Hsiao. A voucher specimen (NPTU140530) was deposited in the Herbarium of National Pingtung University, Pingtung, Taiwan.

2. Previous Studies

Maxillaria tenuifolia Lindl., also known as the "coconut orchid", is an orchid belonging to the Epidendroideae subfamily ranging from Mexico to Nicaragua. These plants are easy to grow if the environment keeps moist and provides the good air movement in a high-light field. *Maxillaria tenuifolia* is a longtime favorite plant because of its strong coconut odor. This is a rare property

* Corresponding author: E-Mail: cyl@mail.nptu.edu.tw; Phone: +886-8-7663800 ext. 33256

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because most of the orchids are scentless to a human nose. Previous chemical investigation of *Maxillaria* species discussed the isolation and structure elucidation of several phenanthrene derivatives [1, 2]. Although the floral volatile compounds of *M. tenuifolia* have been investigated by using headspace-solid phase microextraction coupled to gas chromatography and mass spectrometry [3, 4], the non-volatile constituents of *M. tenuifolia* have not been subjected through the phytochemical analysis.

The rapidly increasing diabetes mellitus is becoming a serious threat to mankind's health in all parts of the world. The control and treatment of diabetes and its complications mainly depend on the chemical or biochemical agents, but it has never been reported that someone is recovered totally from diabetes [5]. The α -glucosidase inhibitor is effective in treating diabetes mellitus and widely used in patients with type-2 diabetes. α -Glucosidase is an enzyme of the intestinal brush border such as hydrolyses terminal, non-reducing 1,4-linked α -D-glucose residues and releases α -D-glucose. Inhibition of α -glucosidase diminishes glucose resorption and postprandial hyperglycemia [6]. Acarbose has been shown the significant improvement of glycemic control in type II diabetes patients [7]. Likewise, herbal inhibitors of α -glucosidase could have the potential ability to decrease postprandial hyperglycemia.

In our ongoing investigation of α -glucosidase inhibitors from the natural source, the EtOAc extract of *M. tenuifolia* (flower) showed significant inhibition of α -glucosidase (82.9 % inhibition at 1 mg/mL), indicated that the EtOAc extract of the flower of *M. tenuifolia* may function as an antihyperglycemic. To understand the inhibitive principle more clearly and to develop an antihyperglycemic drug, the large-scale extraction and fractionation of the flower of *M. tenuifolia* were undertaken.

3. Present Study

This paper dealt with the isolation and identification of compounds **1** - **4** from the flower of *Maxillaria tenuifolia* and analyzed the inhibitory activity of the isolates in relation to acarbose towards α -glucosidase. Furthermore, the antioxidative properties of isolated constituents were also discussed.

Equipment: Melting points were measured on a Yanagimoto MP-S3 micro melting point apparatus and are uncorrected. The UV spectra were recorded on a Hitachi U-3900 spectrophotometer in MeOH solution. The FT-IR spectra were measured on an Agilent Technologies Cary-630 spectrophotometer as KBr disks. The $^1\text{H-NMR}$ (400 MHz) and $^{13}\text{C-NMR}$ (100 MHz) spectra were recorded on a Varian-400 Unity Plus spectrometer. Chemical shifts are shown in δ values with tetramethylsilane as an internal reference. α -Glucosidase (EC 3.2.1.20) from Baker's yeast (*Saccharomyces cerevisiae*) and *p*-nitrophenyl α -D-glucopyranoside (*p*NPG) as a synthetic substrate were purchased from Sigma-Aldrich (St. Louis, MO, USA). Acarbose (Glucobay®; Bayer Vital, Leverkusen, Germany) as a synthetic inhibitor of α -glucosidase was obtained from a local pharmacy.

Extraction and Separation: The fresh flowers of *Maxillaria tenuifolia* (2.8 kg) were soaked with EtOAc (5 L \times 3) at room temperature, and the combined extracts were concentrated under reduced pressure to give a deep brown syrup (**MTE**, 47.1 g). The extract was directly chromatographed on a silica gel column by elution with a gradient of $\text{CHCl}_3/\text{MeOH}$ to afford ten fractions. Fr. 3 underwent column chromatographic separation over silica gel using $\text{CHCl}_3/\text{Me}_2\text{CO}$ (19:1) as an eluent to yield 3,4-dihydroxy benzoic acid methyl ester (**1**) (20 mg) and flavanthridin (**2**) (91 mg). Fr. 9 was rechromatographed on a silica gel column and eluted with $\text{CHCl}_3/\text{MeOH}$ (9:1) to give vanillic acid (**3**) (443 mg). Fr. 10 was recrystallization by EtOAc/MeOH (2:1) to afford mangiferin (**4**) (14.6 g).

Assay for α -Glucosidase Inhibitory Activity: Based on the chromogenic method described by Kim *et al.* [8], we made slight modifications for the α -glucosidase inhibitory assay by using α -glucosidase from Baker's yeast. The substrate solution *p*-nitrophenyl α -D-glucopyranoside was prepared in 0.1 M phosphate buffer, adjusted to pH 6.9, to simulate a model of intestinal fluid. Briefly, α -glucosidase was dissolved in 0.1 M phosphate buffer, pH 6.9, to yield a 1 U/mL stock-solution and pre-incubated

for 10 min at 37 °C with 100 μ L of the respective test solution (DMSO solutions of extracts or compounds at 1.0 mg/mL concentration). The enzymatic reaction was initiated by adding 0.95 mM *p*NPG and the reaction mixture was incubated for another 20 min at 37 °C. The catalytic reaction was terminated by addition of a 1 M Na_2CO_3 solution. Activity of α -glucosidase was determined by measuring the product *p*-nitrophenol released from *p*NPG at 405 nm using a Synegy 2 Microplate Reader from Biotek.

Assay for Antioxidative Activity: In the present study, the ferric thiocyanate method was routinely used for the measurement of the antioxidative activity according to Chen et al. with a slight modification [9]. For oxidation, 1.0 mL of 0.1 M sodium phosphate buffer (pH 7.0), 0.5 mL of distilled water, and 1.0 mL of 50 mM linoleic acid in ethanol (99.5 %) were mixed in a glass test tube (5 mL volume). Test samples were added with the aforementioned buffer or ethanol by keeping the total volume. The tubes were sealed and kept at 60 °C in the dark. To the reaction mixture (80 μ L) were added 30% ammonium thiocyanate (20 μ L), and 20 mM ferrous chloride solution (20 μ L). After 3 min, the absorbance of the colored solution was measured at 500 nm with a Hitachi U-3900 spectrophotometer.

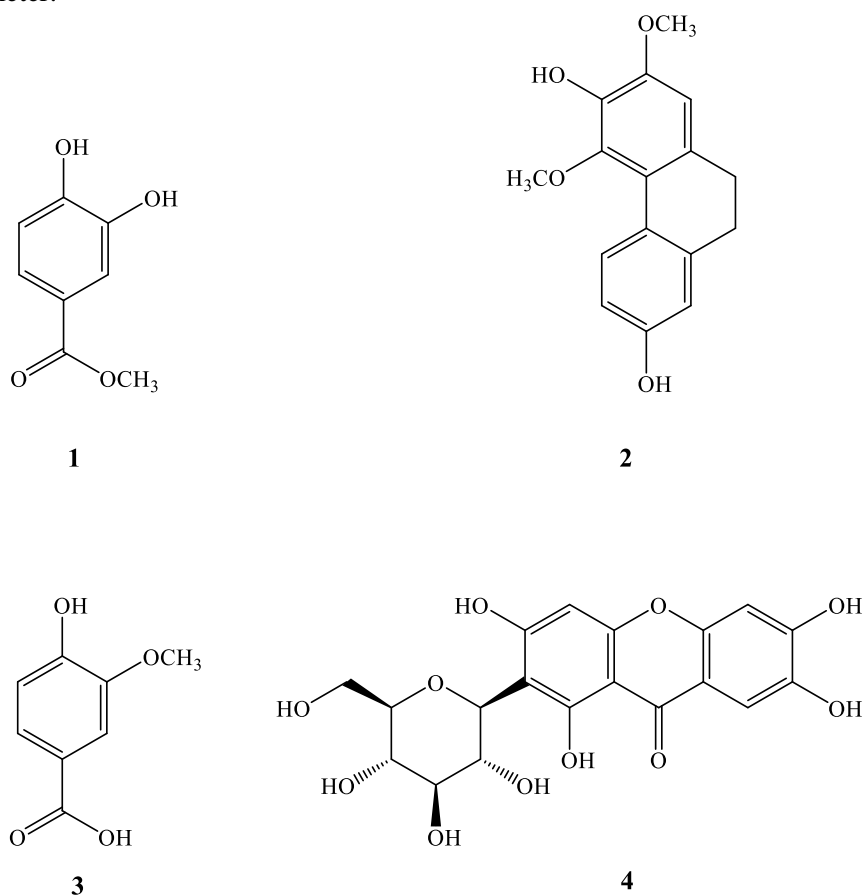


Figure 1. The structures of 1 - 4

Bioassay-Guided Isolation of 1-4: We discovered that the EtOAc extract of *M. tenuifolia* (flower) inhibited α -glucosidase (82.9 % at 1 mg/mL), when tested in an assay for α -glucosidase inhibitory activity. This extract was subjected to α -glucosidase activity-guided fractionation by using column chromatography on silica gel and the activity was limited to the fractions 3, 9 and 10. Further purification processes led to yield four compounds, 3,4-dihydroxy benzoic acid methyl ester (**1**), flavanthridin (**2**), vanillic acid (**3**) and mangiferin (**4**)(Figure 1). All were identified by comparing spectroscopic properties with values previously reported in the literature [10-13]. Among them,

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mangiferin (**4**), a xanthone glucoside, is an active phytochemical present in various plants, with Anacardiaceae and Anemarrhena being two main resources [14]. Mangiferin has been reported to possess antibacterial [13], antitumor [15], antiviral [16] and immunomodulatory activities [17]. Interestingly, in this study, mangiferin (**4**) was the major constituent and approximately 31 % weight of EtOAc extract of *M. tenuifolia*. The other three constituents (**1**), (**2**), and (**3**) have previously been isolated from several plants, but the hypoglycemic properties have not been previously reported.

3,4-Dihydroxy benzoic acid methyl ester (1): White powder. UV λ max (MeOH) nm: 224, 275. IR ν (KBr) cm^{-1} : 1685, 1604, 1521, 1438, 1301. $^1\text{H-NMR}$ (400 MHz, Acetone- d_6): δ 9.83 (1H, s, OH), 8.62 (1H, s, OH), 7.46 (1H, dd, $J = 7.8, 2.0$ Hz, H-6), 7.45 (1H, d, $J = 2.0$ Hz, H-2), 7.02 (1H, d, $J = 7.8$ Hz, H-5), 3.94 (3H, s, OCH_3).

Flavanthridin (2): White powder. UV λ max (MeOH) nm: 205 (sh), 210 (sh), 230, 235 (sh). IR ν (KBr) cm^{-1} : 3400, 2910, 1610, 1490, 1400, 1230, 1122, 1076. $^1\text{H-NMR}$ (400 MHz, Acetone- d_6): δ 8.23 (1H, d, $J = 8.8$ Hz, H-4), 8.13 (1H, s, OH), 7.88 (1H, s, OH), 6.69 (2H, m, Ar-H), 6.46 (1H, s, Ar-H), 3.85 (3H, s, OCH_3), 3.79 (3H, s, OCH_3), 2.67 (4H, m, H-9, H-10).

Vanillic acid (3): White powder. UV λ max (MeOH) nm: 203, 225, 289. IR ν (KBr) cm^{-1} : 3400~2500, 1682, 1598, 1526, 1436, 1286. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6): δ 12.51 (1H, br, COOH), 9.86 (1H, s, 4-OH), 7.43 (1H, dd, $J = 8.8, 2.0$ Hz, H-6), 7.42 (1H, d, $J = 2.0$ Hz, H-2), 6.83 (1H, d, $J = 8.8$ Hz, H-5), 3.80 (3H, s, OCH_3).

Mangiferin (4): Pale yellow powder. UV λ max (MeOH) nm: 246, 260, 333, 374. IR ν (KBr) cm^{-1} : 3500~2600, 2910, 1650, 1610, 1490, 1460, 1250, 1195. $^1\text{H-NMR}$ (400 MHz, DMSO- d_6): δ 13.83 (1H, s, 1-OH), 7.32 (1H, s, H-8), 6.79 (1H, s, H-5), 6.34 (1H, s, H-4), 4.57 (1H, d, $J = 9.6$ Hz, H-1'), 4.04 (1H, dd, $J = 9.6, 9.2$ Hz, H-2'), 3.67 (1H, d, $J = 11.2$ Hz, H-6'a), 3.39 (1H, dd, $J = 11.2, 4.8$ Hz, H-6'b), 3.18 (1H, t, $J = 9.2$ Hz, H-3'), 3.12 (2H, m, H-4', H-5'). $^{13}\text{C-NMR}$ (100 MHz, DMSO- d_6): δ 179.1 (C-9), 164.0 (C-3), 162.0 (C-1), 156.4 (C-4a), 155.6 (C-6), 151.3 (C-10a), 144.3 (C-7), 111.2 (C-8a), 107.8 (C-8), 107.6 (C-2), 102.5 (C-5), 101.4 (C-9a), 93.5 (C-4), 81.8 (C-5'), 79.2 (C-3'), 73.3 (C-1'), 70.9 (C-4'), 70.4 (C-2'), 61.7 (C-6').

α -Glucosidase Inhibitory Activity: The inhibitory activities of the EtOAc extract of *M. tenuifolia* (MTE) in comparison with that of acarbose, the synthetic inhibitor of α -glucosidase, as a reference compound were obtained (Figure 2). The inhibitory activity of the MTE was higher than that of acarbose with 82.9% inhibition. Since MTE displayed significant inhibition of α -glucosidase activity, we analyzed which constituent from the extract was effective. At 1.0 mg/mL concentration, the four isolates **1-4** exhibited 22.9%, 96.7%, 81.0% and 59.9% inhibitory activity towards α -glucosidase, respectively. In this assay flavanthridin (**2**) exhibited the most potent inhibitory activity of all tested compounds. Mangiferin (**4**), the major constituent of MTE, also showed the stronger activity of inhibition from α -glucosidase. Notably, at 1.0 mg/mL concentration, acarbose had a lower ability to inhibit α -glucosidase from *Saccharomyces cerevisiae* than MTE, **2**, **3** and **4**.

Assay for Antioxidative Activity: Anti-oxidant activity of MTE and the isolates **1-4** were determined in ferric thiocyanate method. In this assay, the inhibitory activity of linoleic acid peroxidation was determined and the antioxidant trolox (a water-soluble analogue of tocopherol) was assayed as a reference compound. Unsaturated fatty acids such as linoleic acid will be peroxidized over time to hydroperoxides. In the presence of ferrous, lipid hydroperoxides convert ferrous iron to ferric form, and ferric ion subsequently coordinates with thiocyanate to generate ferric thiocyanate, increasing the absorption at 500 nm. In our studies, 3,4-dihydroxy benzoic acid methyl ester (**1**) and mangiferin (**4**) showed total inhibition on the lipid peroxidation of the linoleic acid emulsion (1.0 mg/mL). In comparison, flavanthridin (**2**) and vanillic acid (**3**) exhibited much lower antioxidant activity (10.5% and 15.5%, respectively) (Figure 3). On the other hand, trolox (standard antioxidant) indicated an inhibition of 76.3% on peroxidation of linoleic acid emulsion.

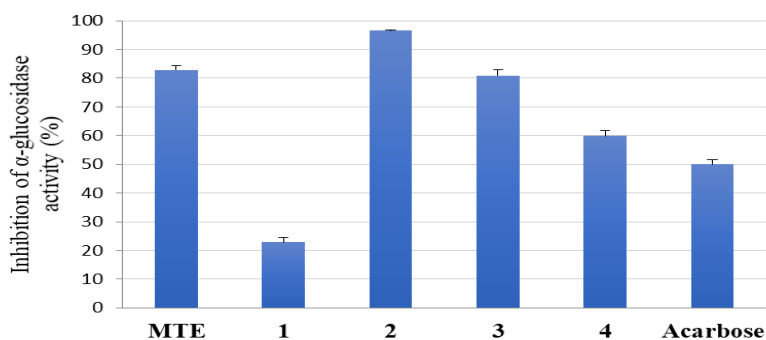


Figure 2. Inhibition of α -glucosidase by 1 mg/mL of EtOAc extract of *M. tenuifolia* (MTE) and constituents **1-4**. Acarbose acts as a positive control. The columns represent mean and standard deviation of three experiments

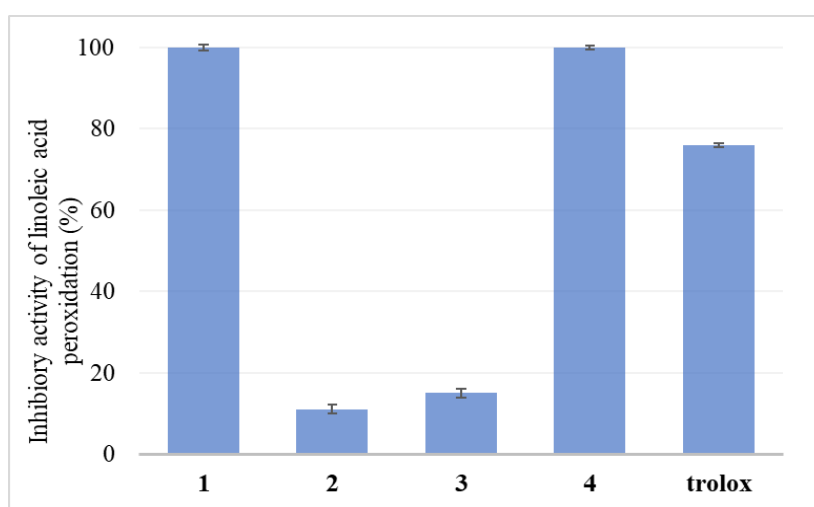


Figure 3. Inhibition of linoleic acid peroxidation by 1 mg/mL of constituents **1-4**. Trolox acts as a positive control. The columns represent mean and standard deviation of three experiments

In conclusion, the EtOAc extract of *M. tenuifolia* (flower) is an effective inhibitor of α -glucosidase and may function as an antihyperglycemic. To understand the inhibitive principle more clearly and to develop an antihyperglycemic drug, we purified and identified the active components from the extract and screened the α -glucosidase inhibitive activity and antioxidative property. The results showed that the EtOAc extract of *M. tenuifolia* and the related constituents might have an anti-diabetic effect by suppressing carbohydrate disintegration. Besides, mangiferin was reported for the first time from *M. tenuifolia* as a major constituent, indicated that *M. tenuifolia* can be used as a new source plant for mangiferin. This study also points out the potential use of *M. tenuifolia* extracting components could be antioxidants for health supplements and preventing damage to organisms by oxidative stress.

Acknowledgments

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Supporting Information

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Anti-diabetic activity of *M. tenuifolia*Chia-Ying Li: [0000-0002-8319-5768](https://orcid.org/0000-0002-8319-5768)

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