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Aspilactonol J: A New Cytotoxic Polyketide Isolated from the

Marine-Derived Fungus Aspergillus sp. w2-13

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Abstract: Marine-derived Aspergillus sp. strain w2-13, isolated from Dongshan Island in Fujian Province, produced five compounds through solid-state fermentation. This included a new polyketide, aspilactonol J (1), and four previously identified compounds (2–5). The structure of the new compound was determined using NMR, HR-MS, and ECD, supported by theoretical calculations. A plausible biosynthetic pathway for polyketides 1–3 was also proposed. Cytotoxicity tests on various cancer cell lines revealed that aspilactonol J (1) exhibited significant selectivity and efficacy, especially against HepG2 cells, suggesting its potential for pharmaceutical development. This study highlights the role of marine fungi as valuable sources of bioactive natural products with therapeutic potential.

Keywords: *Aspergillus*; Polyketide; NMR; ECD; TDDFT; cytotoxic. © 2024 ACG Publications. All rights reserved.

1. Microorganism Source

The strain w2-13 was isolated from a seawater sample collected near Dongshan Island, Fujian Province, China. Its ITS rDNA sequence has been submitted to the NCBI database (accession number: PP873196). A BLAST analysis revealed that the sequence of strain w2-13 exhibits a 100% similarity to that of *Aspergillus ochraceopetaliformis* strain MF4 (accession number: MH141440.1).

2. Previous Studies

Marine natural products often display unique structures and potent biological activities not found in their terrestrial counterparts, making them invaluable for novel drug discovery [1-3]. Fungi

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A new Polyketide from Aspergillus sp.

from the genus *Aspergillus* are prolific sources of natural products. *Aspergillus* are widespread not only in terrestrial ecosystems but also in marine environments, where they have uniquely adapted to saline conditions [4, 5]. *Aspergillus* species are recognized for their ability to synthesize a diverse array of secondary metabolites, several of which exhibit notable biological properties [6-9]. Consequently, this genus has become a central focus in natural product research, yielding compounds with antitumor, antimicrobial, and anti-inflammatory properties [10]. Exploring *Aspergillus* species from marine sources thus offers a promising pathway for discovering novel natural products, highlighting the importance of these fungi in the search for new therapeutic agents [11].

3. Present Study

In this study, a new polyketide (1), along with two known polyketides (2-3) and two known sesquiterpenoid nitrobenzoyl esters (4-5) (Figure 1) were isolated from the marine-derived fungus *Aspergillus* sp. w2-13.

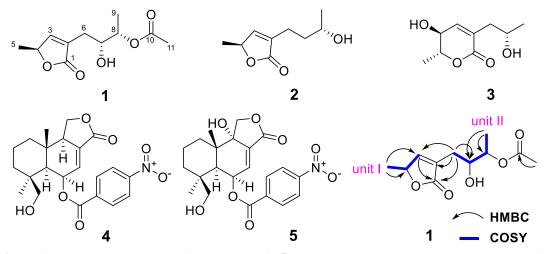


Figure 1. Chemical structures of compounds 1-5 and key COSY and HMBC correlations of 1

Aspergillus sp. strain w2-13 was subjected to solid-state static fermentation using 150 1L Erlenmeyer flasks, each containing 70 g of rice, 30 g of millet, and 100 mL of seawater. After inoculation, the cultures underwent a 25-day incubation at 28°C. Subsequently, each flask underwent three rounds of extraction using 500 mL of ethyl acetate. The resulting extracts were concentrated under vacuo until dry, ultimately producing 117 g of crude extract. This crude extract was subjected to column chromatography on silica gel, eluting with a chloroform-methanol gradient (100:0 to 1:1). The eluent was fractionated into 70 separate 1 L portions. Based on thin-layer chromatography (TLC) analysis, similar components were combined as follows: fractions 1-13 into Group A, 14-21 into Group B, 22-34 into Group C, and 35-39 into Group D. Group B underwent decolorization using gel filtration chromatography (Sephadex LH-20, methanol). Subsequently, it was separated using reversed-phase silica gel column chromatography, employing a methanol-water gradient (100:0 to 0:100), followed by TLC to combine similar fractions, resulting in four subfractions (B1-B4). The B2 subfraction was further purified by medium-pressure liquid chromatography (MPLC) with a 10-50% acetonitrile-water gradient, yielding compound 3 (1.8 mg, $t_{\rm R}$ = 16.1 min). The B4 subfraction was separated using gel filtration chromatography (Sephadex LH-20, methanol), producing B4-1 and B4-2. B4-1 was further separated by MPLC using 50% acetonitrile in water, resulting in compound 5 (2.3 mg, $t_{\rm R}$ = 15.2 min). B4-2 was subjected to MPLC with 30% acetonitrile in water, resulting in the isolation of compound 4 (2.2 mg, $t_{\rm R}$ = 9.9 min). Group C was fractionated into C1-C6 using gel filtration chromatography (Sephadex LH-20, methanol). The C2 fraction was further subjected to separation using silica gel column chromatography with chloroform-methanol (10:1), producing two subfractions, C2-1 and C2-2. C2-2 was purified using MPLC with 45% acetonitrile in water, yielding compound 1 (2.5 mg, $t_{\rm R} = 19.3$

min). C3 was decolorized using gel filtration (Sephadex LH-20, methanol) and then separated on a silica gel column with chloroform-methanol (15:1) as the eluent. Following TLC analysis, compound 2 (2.0 mg) was isolated.

Aspilactonol J (*I*): A white amorphous powder; $[\alpha]_{25}^{D}$ +33.6 (*c* 0.05, MeOH); UV (MeOH) λ_{max} (log ε): 213 (3.83) nm; ECD (*c* 1 mg/mL, MeOH) λ_{max} ($\Delta\varepsilon$): 212 (4.18), 247 (-2.23) nm; HR-ESI-MS *m/z* 251.0901 [M + Na]⁺ (calcd for C₁₁H₁₆O₅Na, 251.0895); IR (KBr) v_{max} 3467, 2961, 2919, 2851, 1733, 1456, 1374, 1321, 1259, 1084, 1050, 1027, 955, 857, 801 cm⁻¹ (Figure S10); ¹H and ¹³C NMR data, Table 1.

Cytotoxic Assay: Cancer cell lines MCF7, HepG2, and L-02 were sourced from Wuhan Pricella Biotechnology Co., Ltd. (Wuhan, China). The cell lines were cultivated following established protocols. To assess the effects of compounds on cancer cell viability, CCK8 assays were conducted according to methods previously described [12].

NMR and ECD Calculations: The use of the CREST tool [13, 14] facilitated the generation of conformer candidates, and DFT calculations were executed employing the Gaussian 16 software. A comprehensive optimization and analysis of molecular conformers was performed, with details provided in the supplementary materials (Appendix S1).

NMR shielding constants were computed and converted to chemical shifts relative to TMS, and DP4+ probabilities were determined. Key statistical parameters such as R^2 , the mean absolute error (MAE), and the corrected mean absolute error (CMAE) were also calculated. Detailed computational procedures and data are provided in Appendix S1.

Calculations using time-dependent density-functional theory (TDDFT) were performed with the CAM-B3LYP/6-311G(d) level in a methanol environment, utilizing the IEFPCM solvent model. For each conformer, 36 excited states were determined [15], and ECD spectra generation was carried out employing Multiwfn 3.6 software [16].

No.	Atom Type	δ _H (Multiplicity, J, nH)	δς
1	С	-	174.4
2	С	-	130.9
3	CH	7.42 (d, 1.5 Hz, 1H)	153.1
4	CH	5.07 (qd, 6.8,1.7 Hz, 1H)	78.4
5	CH_3	1.37 (d, 6.8 Hz, 3H)	19.3
6a	CH_2	2.47 (ddt, 15.1, 3.6, 1.7 Hz, 1H)	29.7
6b	CH_2	2.35 (ddt, 15.1,8.8,1.4 Hz, 1H)	29.7
7	CH	3.87 (m, 1H)	71.7
8	CH	4.78 (qd, 6.4,5.0 Hz, 1H)	73.7
9	CH_3	1.23 (d, 6.4 Hz, 3H)	15.5
10	С	-	170.5
11	CH ₃	1.99 (s, 3H)	21.2

Table 1. ¹H (600 MHz) and ¹³C (150 MHz) NMR data for compound **1** (in acetone- d_6 , δ in ppm, J in Hz)

The molecular formula of **1** was determined to be $C_{11}H_{16}O_5$ based on the sodium adduct ion at m/z 251.0901 (Figure S8), indicating four degrees of unsaturation. Detailed spectroscopic studies, such as ¹H NMR, ¹³C NMR, HSQC, and DEPT-135 analyses (Figures S1-S4), identified specific structural features that are detailed in Table 1 and Table S24. These included two ester carbonyls (δ_C 174.4, 170.5), one sp³ methylene ($\delta_{C/H}$ 29.7/2.47 & 2.35), one olefinic sp² methine ($\delta_{C/H}$ 153.1/7.42), three oxygenated sp³ methines ($\delta_{C/H}$ 78.4/5.07, 71.7/3.87, 73.7/4.78), one olefinic nonprotonated carbon (δ_C 130.9), and three methyl groups ($\delta_{C/H}$ 19.3/1.37, 15.5/1.23, 21.2/1.99). COSY correlation analysis (Figure S6) revealed two spin systems: C-3/C-4/C-5 (unit I) and C-6/C-7/C-8/C-9 (unit II). HMBC correlations of H₃-5 to C-3 and H-3 to C-1/C-2 (Figure S5), along with COSY correlations, confirmed the presence of an $\alpha_{\beta}\beta$ -unsaturated γ -lactone moiety identified as 5-methylfuran-2(5*H*)-one in unit I.

HMBC correlations of H₂-6 to C-1/C-2/C-3 indicated that unit II was attached to C-2 (Figure 1). Chemical shifts ($\delta_{C/H}$ 71.7/3.87, 73.7/4.78) suggested that C-7 and C-8 were oxygenated. HMBC correlations of H8/H₃-11 to C-10 revealed that C-8 was linked to an acetoxy group. The derived structure comprised 11 carbons, 15 hydrogens, and 5 oxygens, confirming the presence of a hydroxy group attached to C-7.

The molecule featured three chiral carbons at C-4, C-7, and C-8. The absolute configuration at C-4 was assigned as *S* based on the positive $\pi \rightarrow \pi^*$ Cotton effect at 210 nm and the negative $n \rightarrow \pi^*$ Cotton effect at 240 nm. The coupling constant between H-7 and H-8 was 5.0 Hz, indicating an *erythro* configuration for the 7,8-diol. Two isomers, (4S,7S,8R)-1 (1a) and (4S,7R,8S)-1 (1b), were analyzed using NMR calculation at the mPW1PW91/6-31+G(d,p) level (Tables S1-S2). The isomer 1b closely matched experimental data as indicated by parameters including R^2 , MAE, and CMAE (Table S3), with a DP4+ probability of 86.98% (Figure S11, Table S3), excluding the possibility of 1a. ECD calculations employing TDDFT at the CAM-B3LYP/6-311G(d) level (Tables S4-S23) corroborated these results, determining the absolute configuration of compound 1 to be 4S,7R,8S. (Figure 2, Figure S9).

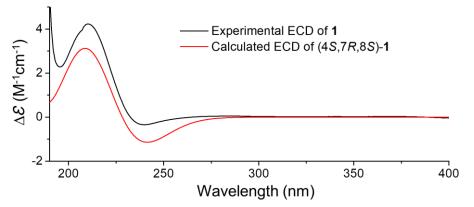


Figure 2. Experimental ECD curve of 1 and calculated ECD spectra of 1b at CAM-B3LYP/6-311G(d)//B3LYP/6-31G(d)

Comparative analysis of NMR data with literature enabled the identification of the remaining four known compounds. These were aspilactonol (2) [17], dihydroaspyrone (3) [17], 14-hydroxy-6 β -p-nitrobenzoylcinnamolide (4) [18] and 9 α ,14-hydroxy-6 β -p-nitrobenzoylcinnamolide (5) [18].

The biosynthesis of compounds 1, 2, and 3 was initiated with the condensation of five C2 units via the polyketide synthase (PKS) pathway to form intermediate **a**. It was oxidized to generate **b**, which underwent molecular rearrangement to produce **c**. Further transformations of **c** through dehydration and decarboxylation led to **d**. Subsequent oxidation of **d** formed **e**, which underwent further oxidation to produce **f**. The transformations of **f** included lactonization to produce **h**, which upon reduction yielded compound 3. Reduction and rearrangement of **f** formed **g**, and subsequent lactonization resulted in compound 2. Meanwhile, reduction, hydration, and rearrangement of **f** produced **i**, which was converted into compound 1 through lactonization and acetylation (Figure 3).

At 20 μ M, compound **1** selectively inhibited the three tested cancer cell lines, achieving a 63.88% inhibition rate in HepG2 cells. In contrast, the positive control drug, doxorubicin (Dox), demonstrated a non-selective inhibition rate exceeding 50% for all tested strains (Table 2). Compounds **2** and **3** displayed significantly lower inhibition rates, below 20% for all strains. IC₅₀ values showed compound **1** at 17.1 μ M and Dox at 73.0 nM on HepG2 cells (Figure S12). These findings indicate that compound **1** possesses a relatively moderate potency in inhibiting HepG2 cells.

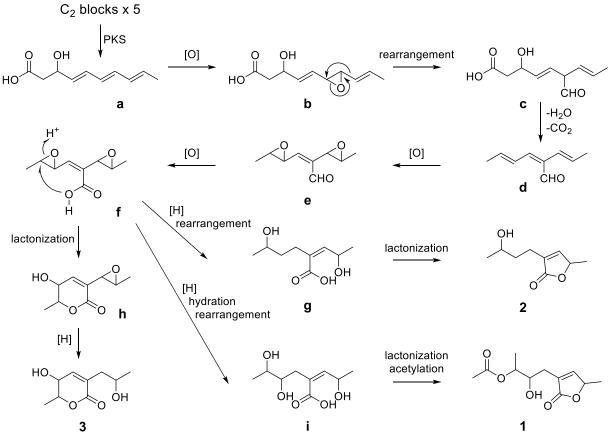


Figure 3. A plausible biosynthetic pathway for compounds 1–3

Table 2. Inhibition rates of compounds 1–3 against three cancer cell lines

	e		
Concentration	Cell inhibition rates \pm SD (%)		
(µM)	MCF-7	HepG2	L-02
20	-2.46 ± 2.86 %	63.88 ± 3.81 %	1.21 ± 6.31 %
20	0.42 ± 2.90 %	1.95 ± 4.64 %	-0.92 ± 1.84 %
20	18.37 ± 2.01 %	4.63 ± 4.32 %	2.82 ± 1.67 %
10	51.48 ± 0.53 %	67.92 ± 1.38 %	$98.42 \pm 0.30~\%$
	(μM) 20 20 20 20	$\begin{array}{c c} (\mu M) & MCF-7 \\ \hline 20 & -2.46 \pm 2.86 \ \% \\ 20 & 0.42 \pm 2.90 \ \% \\ 20 & 18.37 \pm 2.01 \ \% \end{array}$	$\begin{array}{ c c c c c c c c c c c c c c c c c c c$

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Supporting Information

Supporting Information accompanies this paper on <u>http://www.acgpubs.org/journal/records-of-natural-products</u>

A new Polyketide from Aspergillus sp.

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