Supporting Information

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Design, preparation and application of a Pirkle-type chiral stationary phase for enantioseparation of some racemic organic acids and molecular dynamics studies

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Table of Contents	Pages
Table S1: Chromatographic conditions on the separation column.	2
Table S2. Chromatographic conditions in the study Figure S1. The chromatogram for 9th fraction at pH 6	2 3
Figure S2. The chromatogram for 10th fraction at different pH 8	3
Figure S3. Integration of MA-R with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)=-2060,965432 Hartree)	4
Figure S4. Integration of MA-S with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)= -2060,9615805 Hartree)	5
Figure S5. Integration of PP-R with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)= -2064,3802923 Hartree)	6
Figure S6. Integration of PP-S with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)= -2064,3845521 Hartree)	7
Figure S7. The ¹ H NMR and ¹³ C NMR spectra of compound 3 Figure S8. The ¹ H NMR and ¹³ C NMR spectra of compound 4	8 9

Table S1. Chromatographic conditions on the separation column.

Tuble 51: Chromatographic conditions on	the separation column.
Flow rate	0.5 mLmin^{-1}
Temperature	25 °C
Detection wavelenght	220 nm
Mobil phase	0.2 M PBS for each pH
Volume of all test solutions	10 mL
Concentrations of all test solutions	3.0 mgmL^{-1} and 5.0 mgmL^{-1} 0.2 M PBS
pH of test solutions	6.0, 7.0 and 8.0
Number of fractions	12
Volume of fractions	3.0 mL

 Table S2. Chromatographic conditions in the study

Total flow rate	0.8 mLmin^{-1}
Injection volume	3.0 μL
Temperature	25 °C
Detection wavelength	220 nm
Backpressure	150 bar
Mobil phases composition for MA and 2-PPA	n-Hexane/2-PrOH/TFA ^a (80:18:2
	v/v/v)
Analytical column for MA	Chiralpak AD-H
Analytical column for 2-PPA	Chiralpak AD-H
Retention time for MA	15 minutes
Retention time for 2-PPA	12 minutes

^a 5 % TFA solution in 2-propanol



Figure S1. The chromatogram for 9th fraction at pH 6



Figure S2. The chromatogram for 10th fraction at different pH 8



Figure S3. Integration of MA-R with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)=-2060,965432 Hartree)



Figure S4. Integration of MA-S with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)= -2060,9615805 Hartree)



Figure S5. Integration of PP-R with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)= -2064,3802923 Hartree)



Figure S6. Integration of PP-S with CSP from docking (upper) and from quantum mechanical calculations (lower), (E(RB3LYP)= -2064,3845521 Hartree)



Figure 7a. NMR spectrum (CDCl₃; 400MHz) of compound (3)



Figure 7b. ¹³C NMR spectrum (CDCl₃; 100MHz) of compound (3)



Figure 8a. 1 H NMR spectrum (CDCl₃; 400MHz) of compound (4)



Figure 8b. ¹³C NMR spectrum (CDCl₃; 100MHz) of compound (4)