

## Supporting Information

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### Efficient synthesis of benzylidene semicarbazones from aromatic aldehydes by urea-hydrogen peroxide (UHP)

Ali Reza Molla Ebrahimlo\*

*Department of Chemistry, Islamic Azad University, Khoy Branch, P.O. Box 58168–44799, Khoy, Iran*

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\* Corresponding author: E-mail: [mollaebrahimlo@yahoo.com](mailto:mollaebrahimlo@yahoo.com)

### ***S1: General Procedure for the Synthesis of 1-benzylidene semicarbazones Derivatives 2a-h***

A mixture of the aldehyde (**1a-h**) (1 mmol), ammonium acetate (1mmol) and UHP (1mmol) dissolved in acetonitrile (5ml) and the reaction mixture was refluxed. The completion of the reaction was monitored by thin layer chromatography (TLC) (ethyl acetate/n-hexane 3:7). After the completion of the reaction, the reaction mixture was left to cool to room temperature and resulting precipitate was collected by suction filtration, then was washed with water, and dried under vacuum. The dried yellow or white solids were recrystallized from ethanol to afford the desired product as yellow or white crystals.

*1-Benzylidenesemicarbazide (2a)*: white crystal, m. p. : 218-220, Lit.<sup>1</sup>: 222°C. IR (cm<sup>-1</sup>): 1663, 1614, 1471, 1395, 1262, 1091,703, 590. <sup>1</sup>H NMR (ppm): 7.41(br s , 1H) , 7.49 (t,*J*=7.4Hz , 2H) , 7.6 (t,*J*=7.4Hz , 1H) , 7.96 (d, *J*=7.2<sub>H</sub> , 2H) , 8.06 (br s , 1H)

*1-(4-Methylbenzylidene) semicarbazide (2b)*: white crystal, m. p. : 230-234. Lit.<sup>1</sup>:234 °C. IR (cm<sup>-1</sup>): 2854 ,1629 , 1461 , 1399 , 1268 , 670. <sup>1</sup> H NMR(ppm): 1.82 (s , 1H) , 2.35 (s , 3H) , 7.29 (d, *J*=8Hz , 2H) , 7.41(br s , 1H) , 7.86 (d, *J*=8.2Hz , 2H) , 8.1(br s , 1H).

*1-(4-Methoxybenzylidene) semicarbazide (2c)*: white crystal, m. p.:208-211. Lit.<sup>1</sup>:210°C. IR(cm<sup>-1</sup>): 3362 , 3208 , 2949 ,2846 , 1701 , 1637 , 1469 , 1402 , 1250 , 1169 , 1086 , 1017 , 758. <sup>1</sup>H NMR(ppm): 3.81 (s , 3H) , 7.01(d, *J*=8.8Hz , 2H) ,7.36 (br s , 1H) , 7.98 (d, *J*=8.8 , 2H) , 8.12 (br s , 1H) , 10.38 (br s , 1H).

*1-(4-chlorobenzylidene) semicarbazide (2d)*: white crystal, m. p.: 231-233. Lit.<sup>1</sup>: 233 °C. IR(cm<sup>-1</sup>): 3350 , 3219 , 1662 , 1614 , 1467 , 1397 , 1259 , 1088 , 666. HNMR(ppm): 7.44 (br s , 1H) , 7.56 (d, *J*=8.5Hz , 2H) , 7.96 (d, *J*=8.6Hz , 2H) , 7.99 (br s , 1H) , 10.63( s , 1H).

### **References**

- [1] Shriner, Ralph L., Hermann, Christine K. F., Morrill, Terence C., Curtin, David Y., Fuson, Reynold C., The Systematic Identification of Organic Compounds, 8th Edition, John Wiley & Sons, NewYork, **2004**.

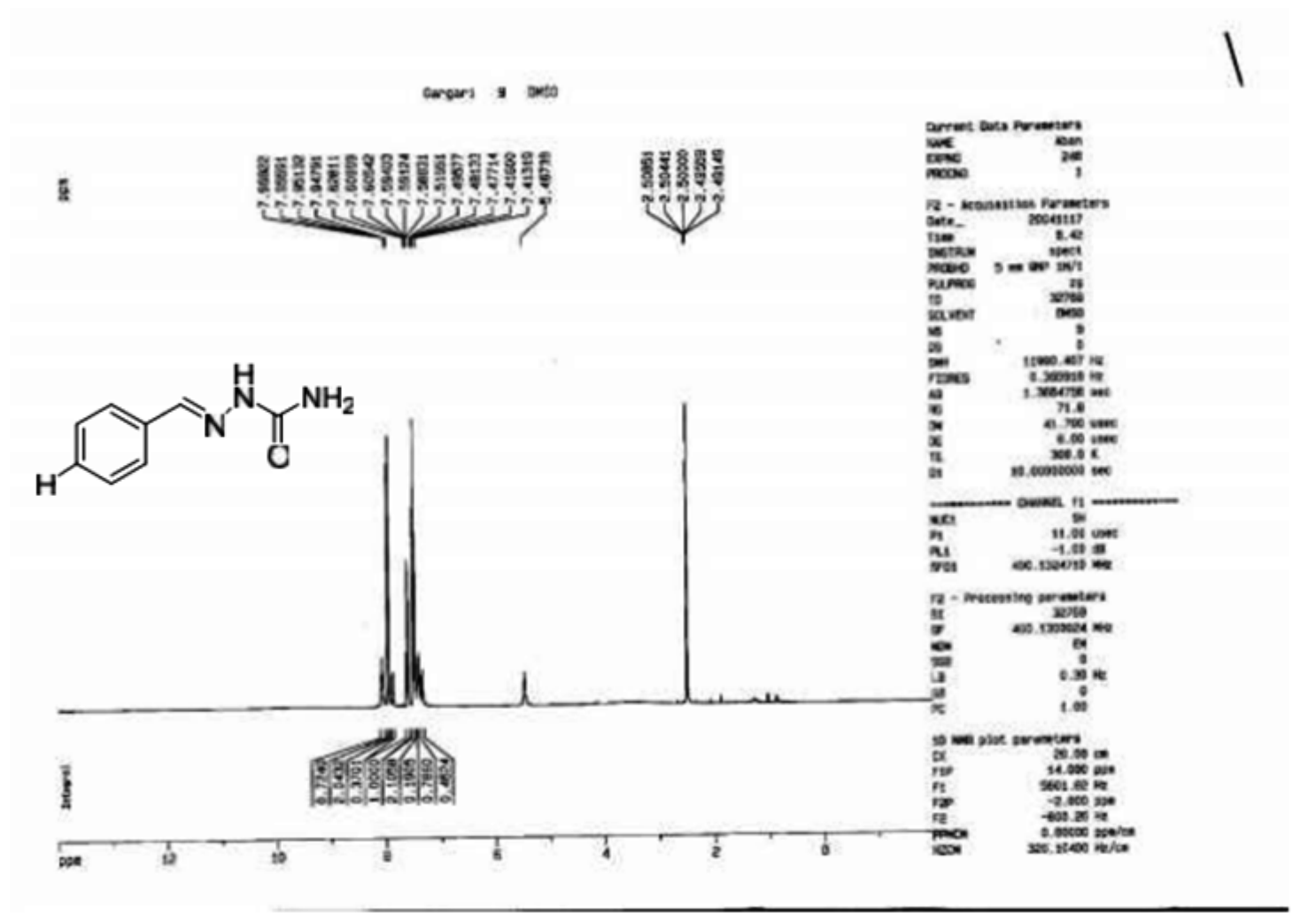
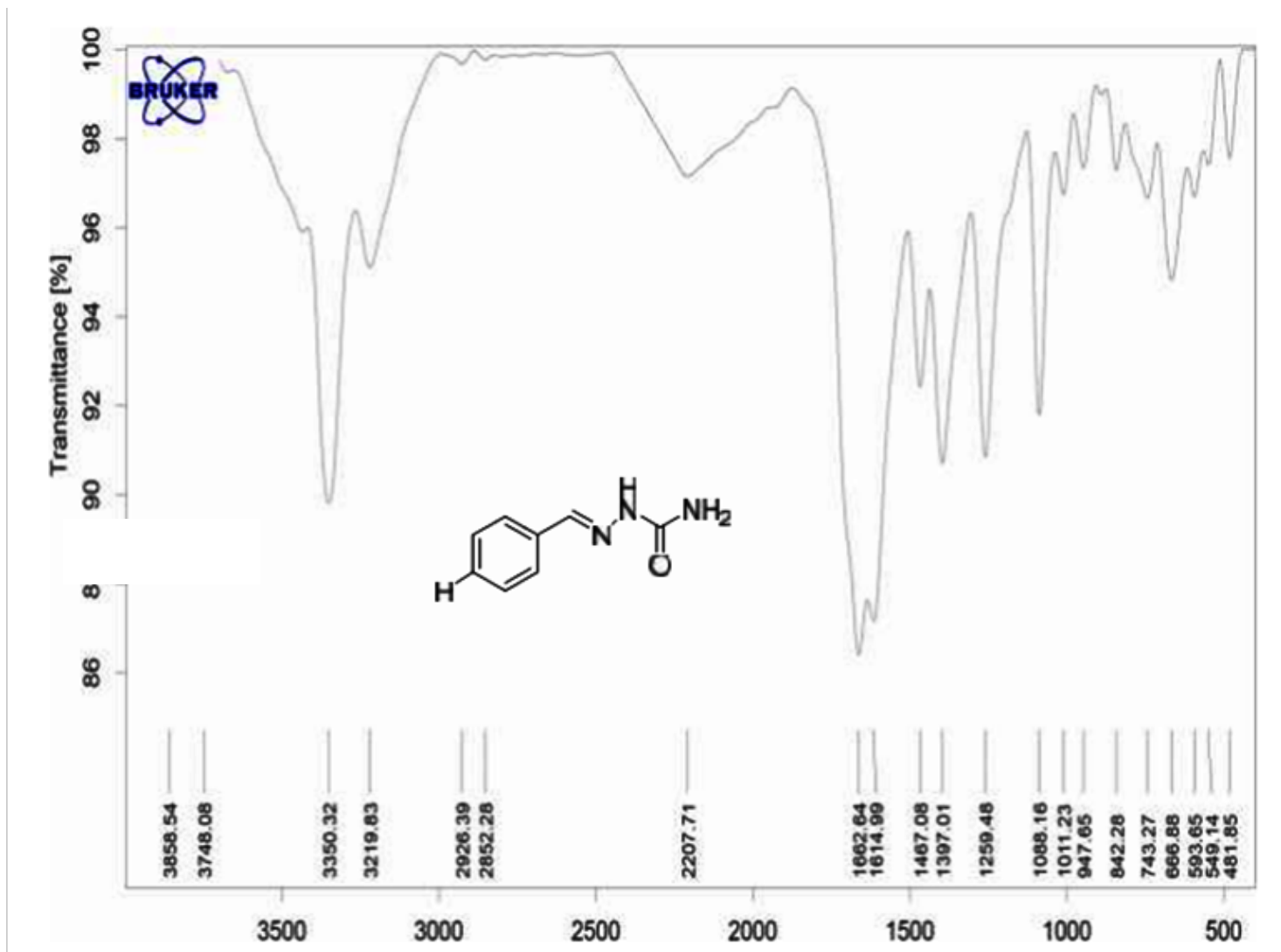


Figure S1 :  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz) spectrum of compound 2a

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**Figure S2** : FT-IR spectrum of compound 2a  
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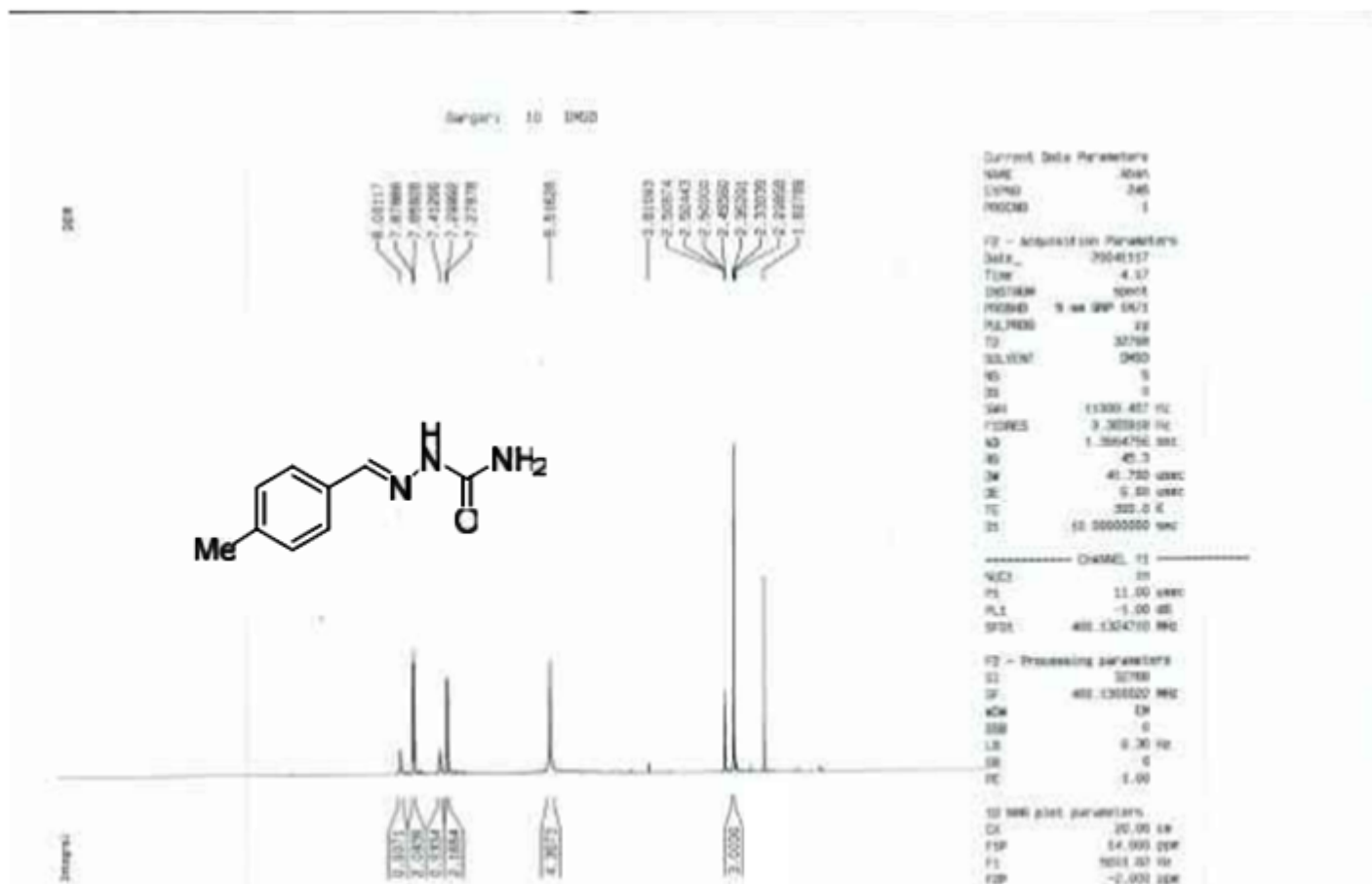


Figure S3 :  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz) spectrum of compound **2b**

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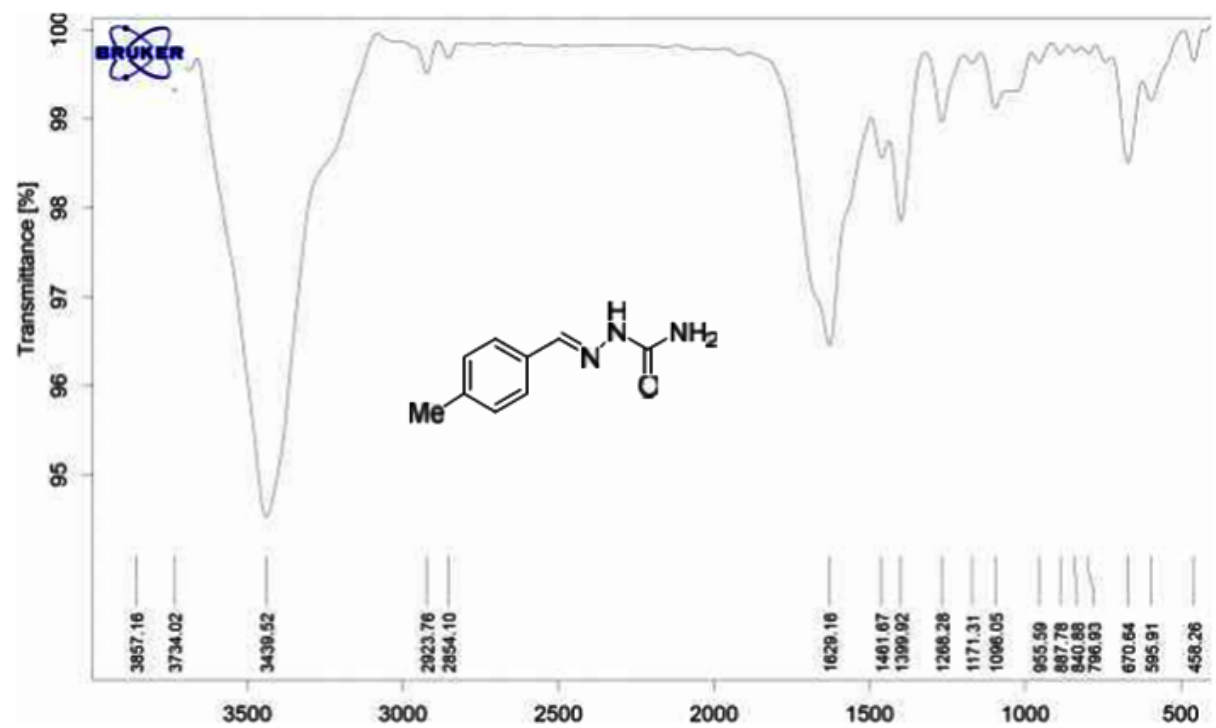


Figure S4 : FT-IR spectrum of compound 2b

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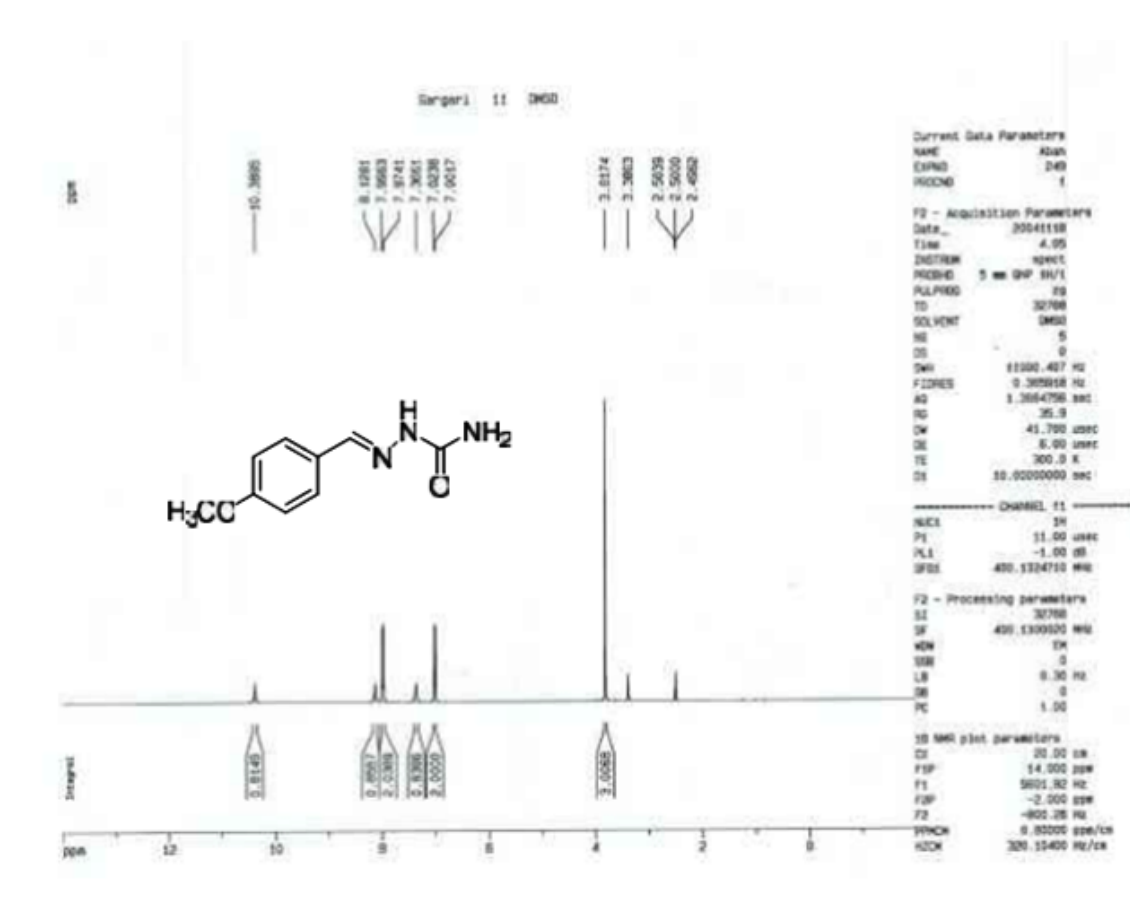


Figure S5 :  $^1\text{H}$  NMR ( $\text{DMSO-}d_6$ , 400 MHz) spectrum of compound **2c**

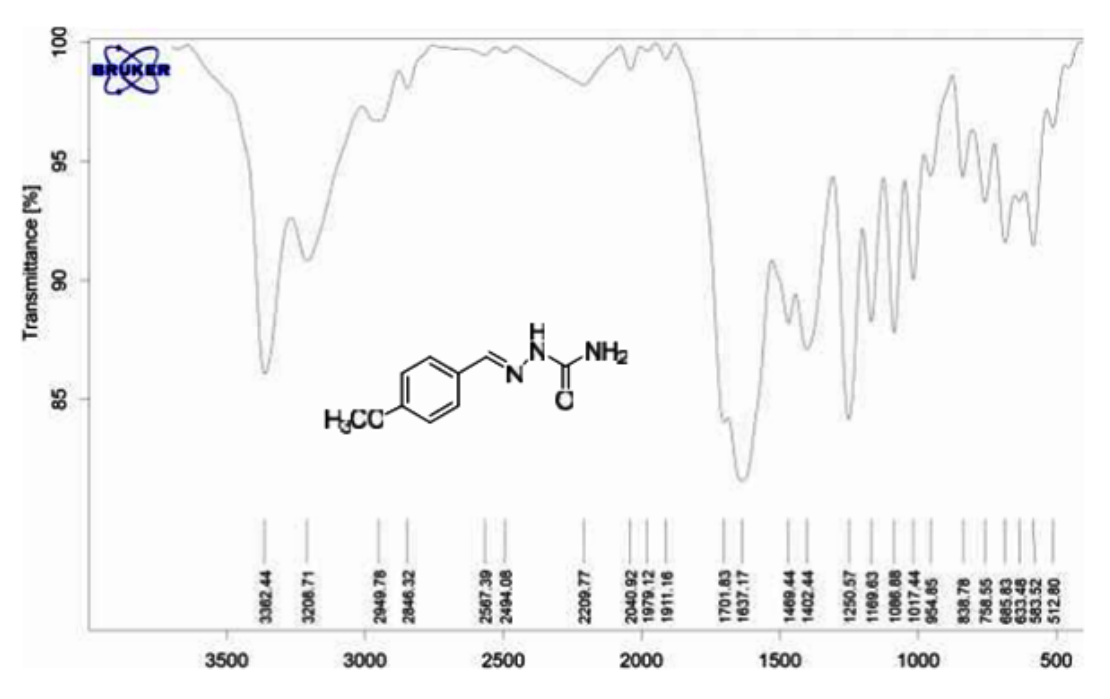
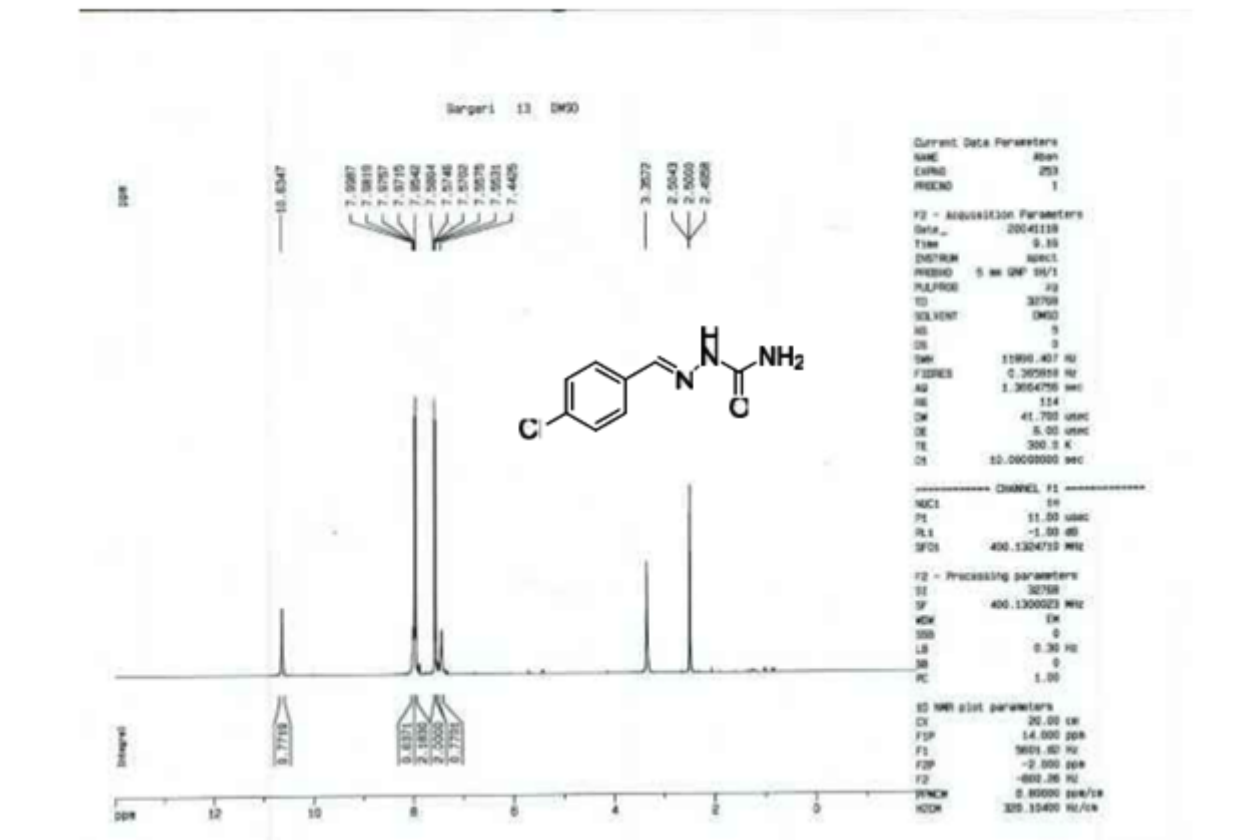


Figure S6 : FT-IR spectrum of compound 2c





**Figure S7** :  $^1\text{H}$  NMR (DMSO- $d_6$ , 400 MHz) spectrum of compound **2d**  
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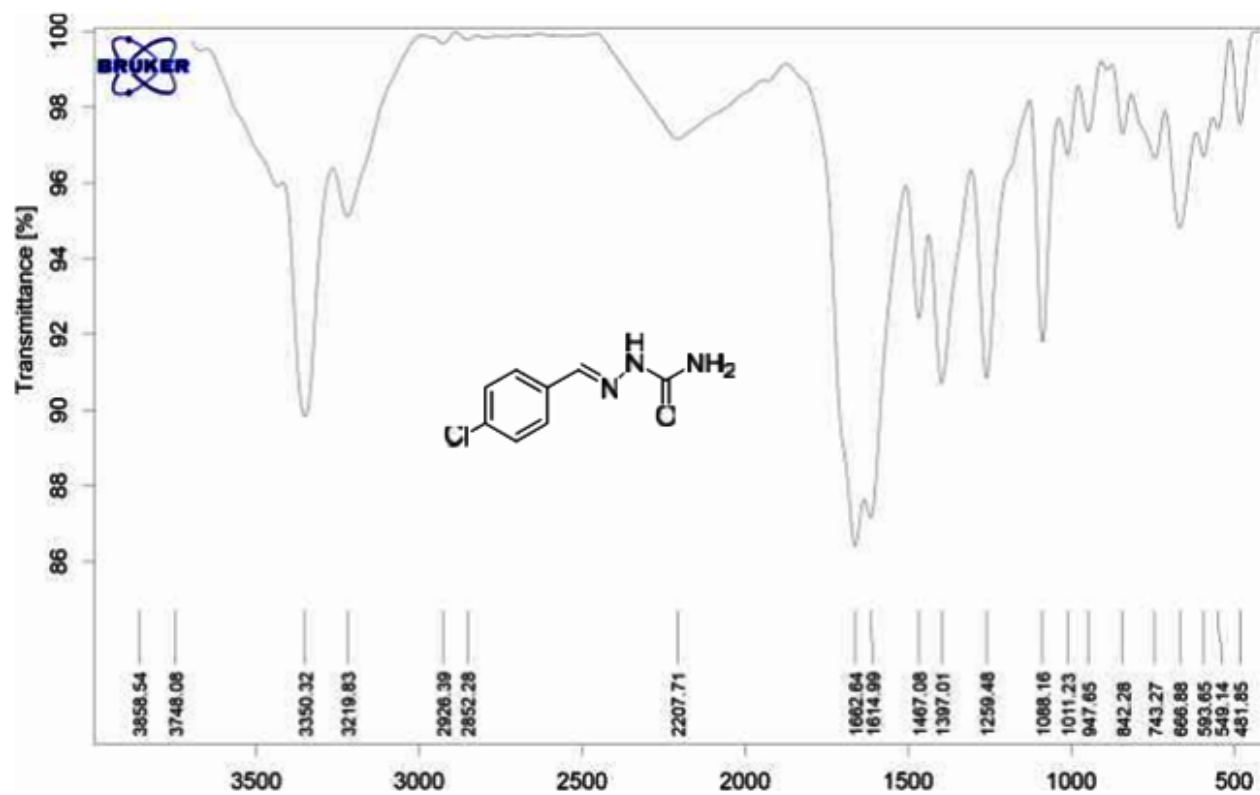


Figure S8 : FT-IR spectrum of compound 2d

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